

# Benzoic acid, 3-(1-methylpropyl)oxy-, 1-methylpropyl ester

<b>Inchi:</b>	InChI=1S/C15H22O3/c1-5-11(3)17-14-9-7-8-13(10-14)15(16)18-12(4)6-2/h7-12H,5-6H2,
<b>InchiKey:</b>	JCGDBWWJKRWEPF-UHFFFAOYSA-N
<b>Formula:</b>	C15H22O3
<b>SMILES:</b>	CCC(C)OC(=O)c1cccc(OC(C)CC)c1
<b>Mol. weight [g/mol]:</b>	250.33

## Physical Properties

Property code	Value	Unit	Source
gf	-165.60	kJ/mol	Joback Method
hf	-515.45	kJ/mol	Joback Method
hfus	25.19	kJ/mol	Joback Method
hvap	62.71	kJ/mol	Joback Method
log10ws	-4.57		Crippen Method
logp	3.819		Crippen Method
mcvol	211.760	ml/mol	McGowan Method
pc	1895.30	kPa	Joback Method
rinpol	1727.00		NIST Webbook
rinpol	1727.00		NIST Webbook
tb	672.09	K	Joback Method
tc	875.92	K	Joback Method
tf	362.14	K	Joback Method
vc	0.797	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	574.95	J/molxK	672.09	Joback Method
cpg	591.84	J/molxK	706.06	Joback Method
cpg	607.76	J/molxK	740.03	Joback Method
cpg	622.73	J/molxK	774.00	Joback Method
cpg	636.74	J/molxK	807.97	Joback Method
cpg	649.82	J/molxK	841.94	Joback Method
cpg	661.98	J/molxK	875.92	Joback Method
dvisc	0.0016383	Paxs	362.14	Joback Method

dvisc	0.0007488	Paxs	413.80	Joback Method
dvisc	0.0004072	Paxs	465.46	Joback Method
dvisc	0.0002501	Paxs	517.12	Joback Method
dvisc	0.0001679	Paxs	568.77	Joback Method
dvisc	0.0001204	Paxs	620.43	Joback Method
dvisc	0.0000909	Paxs	672.09	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U375420&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U375420&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>c</sub>vol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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