

2,3,4-Trifluorobenzoic acid, 4-benzyloxyphenyl ester

Inchi: InChI=1S/C20H13F3O3/c21-17-11-10-16(18(22)19(17)23)20(24)26-15-8-6-14(7-9-15)25
InchiKey: LSPRJYJEPMKXBQ-UHFFFAOYSA-N
Formula: C20H13F3O3
SMILES: O=C(Oc1ccc(OCc2ccccc2)cc1)c1ccc(F)c(F)c1F
Mol. weight [g/mol]: 358.31

Physical Properties

Property code	Value	Unit	Source
gf	-507.12	kJ/mol	Joback Method
hf	-757.77	kJ/mol	Joback Method
hfus	41.34	kJ/mol	Joback Method
hvap	78.70	kJ/mol	Joback Method
log10ws	-6.78		Crippen Method
logp	4.902		Crippen Method
mcvol	240.000	ml/mol	McGowan Method
pc	1872.41	kPa	Joback Method
rinpol	2548.00		NIST Webbook
rinpol	2548.00		NIST Webbook
tb	853.48	K	Joback Method
tc	1082.48	K	Joback Method
tf	540.66	K	Joback Method
vc	0.927	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	689.75	J/molxK	853.48	Joback Method
cpg	702.13	J/molxK	891.65	Joback Method
cpg	713.27	J/molxK	929.81	Joback Method
cpg	723.21	J/molxK	967.98	Joback Method
cpg	731.97	J/molxK	1006.15	Joback Method
cpg	739.60	J/molxK	1044.31	Joback Method
cpg	746.12	J/molxK	1082.48	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U308038&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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