

# 2-Butenoic acid, 3-methyl-, ethyl ester

|                             |  |
|-----------------------------|--|
| <b>Other names:</b>         | 3,3-Dimethylacrylic acid ethyl ester<br>3-Methyl-2-butenoic acid, ethyl ester<br>Crotonic acid, 3-methyl-, ethyl ester<br>Ethyl 3,3-dimethylacrylate<br>Ethyl 3-methyl-2-butenoate<br>Ethyl 3-methylbut-2-enoate<br>Ethyl 3-methylcrotonate<br>Ethyl dimethylacrylate<br>Ethyl isobutenoate<br>Ethyl isopropylideneacetate<br>Ethyl senecioate<br>Ethyl «beta», «beta»-dimethylacrylate<br>Ethyl «beta»-methylcrotonate<br>Ethyl «beta», «beta»-dimethylacrylate<br>Ethyl «beta»-methylcrotonate<br>NSC 61853<br>NSC 99208 |
| <b>Inchi:</b>               | InChI=1S/C7H12O2/c1-4-9-7(8)5-6(2)3/h5H,4H2,1-3H3  |
| <b>InchiKey:</b>            | UTXVCHVLDOLVPC-UHFFFAOYSA-N  |
| <b>Formula:</b>             | C7H12O2  |
| <b>SMILES:</b>              | CCOC(=O)C=C(C)C  |
| <b>Mol. weight [g/mol]:</b> | 128.17   |
| <b>CAS:</b>                 | 638-10-8   |

## Physical Properties

| Property code | Value        | Unit   | Source         |
|---------------|--------------|--------|----------------|
| gf            | -154.19      | kJ/mol | Joback Method  |
| hf            | -325.18      | kJ/mol | Joback Method  |
| hfus          | 15.56        | kJ/mol | Joback Method  |
| hvap          | 49.30 ± 0.20 | kJ/mol | NIST Webbook   |
| log10ws       | -1.47        |        | Crippen Method |
| logp          | 1.516        |        | Crippen Method |
| mcvol         | 112.630      | ml/mol | McGowan Method |
| pc            | 3131.49      | kPa    | Joback Method  |
| rinpol        | 924.00       |        | NIST Webbook   |
| rinpol        | 902.00       |        | NIST Webbook   |
| rinpol        | 924.00       |        | NIST Webbook   |

|        |         |                      |               |
|--------|---------|----------------------|---------------|
| rinpol | 911.00  |                      | NIST Webbook  |
| rinpol | 911.00  |                      | NIST Webbook  |
| rinpol | 911.00  |                      | NIST Webbook  |
| ripol  | 1219.00 |                      | NIST Webbook  |
| ripol  | 1217.00 |                      | NIST Webbook  |
| tb     | 426.70  | K                    | NIST Webbook  |
| tc     | 628.49  | K                    | Joback Method |
| tf     | 221.77  | K                    | Joback Method |
| vc     | 0.432   | m <sup>3</sup> /kmol | Joback Method |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 222.62 | J/mol×K | 439.89          | Joback Method |
| cpg           | 233.55 | J/mol×K | 471.32          | Joback Method |
| cpg           | 244.02 | J/mol×K | 502.76          | Joback Method |
| cpg           | 254.05 | J/mol×K | 534.19          | Joback Method |
| cpg           | 263.64 | J/mol×K | 565.63          | Joback Method |
| cpg           | 272.80 | J/mol×K | 597.06          | Joback Method |
| cpg           | 281.54 | J/mol×K | 628.49          | Joback Method |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.61219e+01                   |
| Coeff. B                    | -4.15272e+03                  |
| Coeff. C                    | -5.97590e+01                  |
| Temperature range (K), min. | 322.02                        |
| Temperature range (K), max. | 443.90                        |

## Sources

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

|   |   |
|---|---|
| <b>Joback Method:</b>                       | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>   |
| <b>McGowan Method:</b>                      | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>   |
| <b>NIST Webbook:</b>                        | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C638108&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C638108&amp;Units=SI</a>   |
| <b>The Yaws Handbook of Vapor Pressure:</b> | <a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> |
| <b>Crippen Method:</b>                      | <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>   |

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>pvap:</b>    | Vapor pressure                                  |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>ripol:</b>   | Polar retention indices                         |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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