

# 4-iodo-2,5-dimethoxy-«beta»-phenethylamine-M, (desamino-HO-), acetylated

InChI: InChI=1S/C13H15IO5/c1-3/15)18-514-10-6-13(17-3)11(14)7-12(10)19-9(2)16/h6-7H,4-5H  
InChIKey: HGOWGSGYCOFNAXUHFFFAOYSA-N

Formula: C13H15IO5

SMILES: COc1cc(CCOC(C)=O)c(OC(C)=O)cc1I

Mol. weight [g/mol]: 378.16

## Physical Properties

Property code	Value	Unit	Source
gf	-372.62	kJ/mol	Joback Method
hf	-654.48	kJ/mol	Joback Method
hfus	33.47	kJ/mol	Joback Method
hvap	78.89	kJ/mol	Joback Method
log10ws	-3.56		Crippen Method
logp	2.331		Crippen Method
mcvol	216.840	ml/mol	McGowan Method
pc	2210.37	kPa	Joback Method
rinpol	2150.00		NIST Webbook
rinpol	2200.00		NIST Webbook
tb	806.60	K	Joback Method
tc	1036.03	K	Joback Method
tf	524.86	K	Joback Method
vc	0.809	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	556.59	J/molxK	806.60	Joback Method
cpg	568.27	J/molxK	844.84	Joback Method
cpg	578.93	J/molxK	883.08	Joback Method
cpg	588.56	J/molxK	921.32	Joback Method
cpg	597.15	J/molxK	959.56	Joback Method
cpg	604.67	J/molxK	997.79	Joback Method
cpg	611.11	J/molxK	1036.03	Joback Method
dvisc	0.0004750	Paxs	524.86	Joback Method

dvisc	0.0003149	Paxs	571.82	Joback Method
dvisc	0.0002221	Paxs	618.77	Joback Method
dvisc	0.0001646	Paxs	665.73	Joback Method
dvisc	0.0001269	Paxs	712.69	Joback Method
dvisc	0.0001011	Paxs	759.64	Joback Method
dvisc	0.0000826	Paxs	806.60	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R514535&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R514535&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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