

# Ethyl 2-cyanoacetoacetate

<b>Other names:</b>	Ethyl $\alpha$ -cyanoacetoacetate Ethyl «alpha»-cyanoacetoacetate
<b>Inchi:</b>	InChI=1S/C7H9NO3/c1-3-11-7(10)6(4-8)5(2)9/h6H,3H2,1-2H3
<b>InchiKey:</b>	NWOKVFOTWMZMHL-UHFFFAOYSA-N
<b>Formula:</b>	C7H9NO3
<b>SMILES:</b>	CCOC(=O)C(C#N)C(C)=O
<b>Mol. weight [g/mol]:</b>	155.15
<b>CAS:</b>	634-55-9

## Physical Properties

Property code	Value	Unit	Source
chs	-3504.00	kJ/mol	NIST Webbook
gf	-224.04	kJ/mol	Joback Method
hf	-385.59	kJ/mol	Joback Method
hfs	-536.80	kJ/mol	NIST Webbook
hfus	16.25	kJ/mol	Joback Method
hvap	57.17	kJ/mol	Joback Method
log10ws	-0.52		Crippen Method
logp	0.278		Crippen Method
mvol	119.880	ml/mol	McGowan Method
pc	3093.29	kPa	Joback Method
tb	591.36	K	Joback Method
tc	798.81	K	Joback Method
tf	340.73	K	Joback Method
vc	0.477	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	277.23	J/mol×K	591.36	Joback Method
cpg	286.29	J/mol×K	625.94	Joback Method
cpg	294.87	J/mol×K	660.51	Joback Method
cpg	302.99	J/mol×K	695.09	Joback Method
cpg	310.64	J/mol×K	729.66	Joback Method

cpg	317.82	J/mol×K	764.24	Joback Method
cpg	324.52	J/mol×K	798.81	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	379.00 ± 1.00	K	2.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C634559&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C634559&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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