

# Bis(3-oxobutan-2-yl) phthalate

<b>Other names:</b>	1,2-Benzenedicarboxylic acid, bis(3-oxobutan-2-yl) ester Bis(3-oxobutan-2-yl)-1,2-benzenedicarboxylate
<b>Inchi:</b>	InChI=1S/C16H18O6/c1-9(17)11(3)21-15(19)13-7-5-6-8-14(13)16(20)22-12(4)10(2)18/h5
<b>InchiKey:</b>	SPKUAVROPJEOQR-UHFFFAOYSA-N
<b>Formula:</b>	C16H18O6
<b>SMILES:</b>	CC(=O)C(C)OC(=O)c1ccccc1C(=O)OC(C)C(C)=O
<b>Mol. weight [g/mol]:</b>	306.31

## Physical Properties

Property code	Value	Unit	Source
gf	-543.94	kJ/mol	Joback Method
hf	-873.83	kJ/mol	Joback Method
hfus	32.57	kJ/mol	Joback Method
hvap	85.18	kJ/mol	Joback Method
log10ws	-3.25		Crippen Method
logp	1.955		Crippen Method
mcvol	230.560	ml/mol	McGowan Method
pc	2058.62	kPa	Joback Method
rinpol	2101.00		NIST Webbook
rinpol	2101.00		NIST Webbook
tb	856.58	K	Joback Method
tc	1076.91	K	Joback Method
tf	523.20	K	Joback Method
vc	0.872	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	677.45	J/mol×K	856.58	Joback Method
cpg	689.35	J/mol×K	893.30	Joback Method
cpg	700.09	J/mol×K	930.02	Joback Method
cpg	709.68	J/mol×K	966.75	Joback Method
cpg	718.12	J/mol×K	1003.47	Joback Method
cpg	725.42	J/mol×K	1040.19	Joback Method

cpg	731.60	J/molxK	1076.91	Joback Method
dvisc	0.0007442	Paxs	523.20	Joback Method
dvisc	0.0004211	Paxs	578.76	Joback Method
dvisc	0.0002632	Paxs	634.33	Joback Method
dvisc	0.0001775	Paxs	689.89	Joback Method
dvisc	0.0001269	Paxs	745.45	Joback Method
dvisc	0.0000951	Paxs	801.02	Joback Method
dvisc	0.0000740	Paxs	856.58	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373499&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373499&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/51-474-9/Bis-3-oxobutan-2-yl-phthalate.pdf>

Generated by Cheméo on 2024-04-30 16:48:36.752758462 +0000 UTC m=+16784965.673335774.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.