

I-Alanine, N-pivaloyl-, methyl ester

Inchi:	InChI=1S/C9H17NO3/c1-6(7(11)13-5)10-8(12)9(2,3)4/h6H,1-5H3,(H,10,12)
InchiKey:	OTUZHONCVOSAC-UHFFFAOYSA-N
Formula:	C9H17NO3
SMILES:	COC(=O)C(C)NC(=O)C(C)(C)C
Mol. weight [g/mol]:	187.24

Physical Properties

Property code	Value	Unit	Source
gf	-248.15	kJ/mol	Joback Method
hf	-547.03	kJ/mol	Joback Method
hfus	17.61	kJ/mol	Joback Method
hvap	56.28	kJ/mol	Joback Method
log10ws	-1.29		Crippen Method
logp	0.710		Crippen Method
mcvol	156.660	ml/mol	McGowan Method
pc	2676.31	kPa	Joback Method
rinpol	1215.00		NIST Webbook
rinpol	1215.00		NIST Webbook
tb	581.98	K	Joback Method
tc	780.42	K	Joback Method
tf	353.36	K	Joback Method
vc	0.588	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	396.49	J/mol×K	581.98	Joback Method
cpg	410.14	J/mol×K	615.05	Joback Method
cpg	423.03	J/mol×K	648.13	Joback Method
cpg	435.19	J/mol×K	681.20	Joback Method
cpg	446.63	J/mol×K	714.27	Joback Method
cpg	457.38	J/mol×K	747.34	Joback Method
cpg	467.46	J/mol×K	780.42	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U299741&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rlnpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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