

# 1,1'-Biphenyl, 3-methyl-

<b>Other names:</b>	(3-Methylphenyl)benzene 1-methyl-3-phenylbenzene 3-Methyl-1,1'-biphenyl 3-Methylbiphenyl 3-Methyldiphenyl 3-Phenyltoluene Biphenyl, 3-methyl- m-Methylbiphenyl m-Phenyltoluene
<b>Inchi:</b>	InChI=1S/C13H12/c1-11-6-5-9-13(10-11)12-7-3-2-4-8-12/h2-10H,1H3
<b>InchiKey:</b>	NPDIDUXTRAITDE-UHFFFAOYSA-N
<b>Formula:</b>	C13H12
<b>SMILES:</b>	<chem>Cc1ccc(-c2ccccc2)c1</chem>
<b>Mol. weight [g/mol]:</b>	168.23
<b>CAS:</b>	643-93-6

## Physical Properties

Property code	Value	Unit	Source
affp	828.00	kJ/mol	NIST Webbook
basg	795.50	kJ/mol	NIST Webbook
chl	-6916.20 ± 7.10	kJ/mol	NIST Webbook
gf	273.77	kJ/mol	Joback Method
hf	149.94	kJ/mol	Joback Method
hfus	17.12	kJ/mol	Joback Method
hvap	69.60	kJ/mol	NIST Webbook
ie	7.95 ± 0.02	eV	NIST Webbook
log10ws	-4.55		Crippen Method
logp	3.662		Crippen Method
mcvol	146.510	ml/mol	McGowan Method
pc	3022.28	kPa	Joback Method
rinpola	1487.70		NIST Webbook
rinpola	1490.00		NIST Webbook
rinpola	1464.00		NIST Webbook
rinpola	1485.00		NIST Webbook
rinpola	255.91		NIST Webbook
rinpola	254.57		NIST Webbook
rinpola	254.70		NIST Webbook

rinpol	1483.50		NIST Webbook
rinpol	256.69		NIST Webbook
rinpol	254.81		NIST Webbook
rinpol	263.90		NIST Webbook
rinpol	254.00		NIST Webbook
rinpol	263.90		NIST Webbook
rinpol	255.91		NIST Webbook
rinpol	254.81		NIST Webbook
rinpol	1474.50		NIST Webbook
rinpol	1487.70		NIST Webbook
rinpol	1464.00		NIST Webbook
rinpol	1487.70		NIST Webbook
rinpol	1475.60		NIST Webbook
rinpol	1483.50		NIST Webbook
rinpol	1464.40		NIST Webbook
rinpol	1474.50		NIST Webbook
rinpol	1480.00		NIST Webbook
rinpol	1478.00		NIST Webbook
rinpol	254.33		NIST Webbook
rinpol	1474.50		NIST Webbook
ripol	2120.00		NIST Webbook
ripol	2104.00		NIST Webbook
ripol	2102.00		NIST Webbook
ripol	2102.00		NIST Webbook
tb	543.00 ± 5.00	K	NIST Webbook
tb	541.00 ± 4.00	K	NIST Webbook
tb	541.00 ± 4.00	K	NIST Webbook
tb	545.85 ± 1.00	K	NIST Webbook
tb	543.00 ± 3.00	K	NIST Webbook
tb	546.00 ± 4.00	K	NIST Webbook
tb	544.00 ± 4.00	K	NIST Webbook
tb	545.90	K	NIST Webbook
tb	548.00 ± 5.00	K	NIST Webbook
tc	800.99	K	Joback Method
tf	277.68 ± 0.10	K	NIST Webbook
tf	296.00 ± 2.00	K	NIST Webbook
vc	0.547	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	407.46	J/molxK	800.99	Joback Method
cpg	396.32	J/molxK	760.03	Joback Method
cpg	384.20	J/molxK	719.06	Joback Method
cpg	371.02	J/molxK	678.09	Joback Method
cpg	356.73	J/molxK	637.12	Joback Method
cpg	341.24	J/molxK	596.15	Joback Method
cpg	324.50	J/molxK	555.18	Joback Method
dvisc	0.0019527	Paxs	301.63	Joback Method
dvisc	0.0001868	Paxs	555.18	Joback Method
dvisc	0.0002351	Paxs	512.92	Joback Method
dvisc	0.0003084	Paxs	470.66	Joback Method
dvisc	0.0004268	Paxs	428.40	Joback Method
dvisc	0.0006341	Paxs	386.15	Joback Method
dvisc	0.0010384	Paxs	343.89	Joback Method
hvapt	68.45	kJ/mol	298.15	Vapour pressures and enthalpies of vaporization of a series of the alkylbiphenyls

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	422.20	K	2.70	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.45143e+01
Coeff. B	-4.46411e+03
Coeff. C	-9.28960e+01
Temperature range (K), min.	406.68
Temperature range (K), max.	577.98

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Vapour pressures and enthalpies of vaporization of a series of the Joback Method:</b>	<a href="https://www.doi.org/10.1016/j.fluid.2012.08.020">https://www.doi.org/10.1016/j.fluid.2012.08.020</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C643936&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C643936&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

# Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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