

# Sulfanilamide, 2,5-dimethoxy-

<b>Other names:</b>	4-amino-2,5-dimethoxybenzenesulphonamide
<b>Inchi:</b>	InChI=1S/C8H12N2O4S/c1-13-6-4-8(15(10,11)12)7(14-2)3-5(6)9/h3-4H,9H2,1-2H3,(H2,
<b>InchiKey:</b>	XUEVFQRBADJXLR-UHFFFAOYSA-N
<b>Formula:</b>	C8H12N2O4S
<b>SMILES:</b>	COc1cc(S(N)(=O)=O)c(OC)cc1N
<b>Mol. weight [g/mol]:</b>	232.26
<b>CAS:</b>	54179-10-1

## Physical Properties

Property code	Value	Unit	Source
gf	-445.64	kJ/mol	Joback Method
hf	-656.54	kJ/mol	Joback Method
hfus	33.50	kJ/mol	Joback Method
hvap	82.40	kJ/mol	Joback Method
log10ws	-1.04		Crippen Method
logp	-0.067		Crippen Method
mcvol	159.610	ml/mol	McGowan Method
pc	4659.34	kPa	Joback Method
tb	661.74	K	Joback Method
tc	882.04	K	Joback Method
tf	493.44	K	Joback Method
vc	0.596	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	406.04	J/molxK	661.74	Joback Method
cpg	417.83	J/molxK	698.46	Joback Method
cpg	428.85	J/molxK	735.17	Joback Method
cpg	439.08	J/molxK	771.89	Joback Method
cpg	448.47	J/molxK	808.60	Joback Method
cpg	456.99	J/molxK	845.32	Joback Method
cpg	464.61	J/molxK	882.04	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C54179101&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C54179101&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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