

# 4-ethyl-2,5-dimethoxy-«beta»-phenethylamine-M, (HO-desamino-COOH), isomer 2, methyl-acetylated

SMILES: CCc1c(OC)cc(CC(=O)OC)c(OC)c1OC(C)=O

Mol. weight [g/mol]: 296.32

## Physical Properties

Property code	Value	Unit	Source
gf	-528.53	kJ/mol	Joback Method
hf	-916.32	kJ/mol	Joback Method
hfus	35.04	kJ/mol	Joback Method
hvap	77.04	kJ/mol	Joback Method
log10ws	-2.91		Crippen Method
logp	1.907		Crippen Method
mcvol	225.070	ml/mol	McGowan Method
pc	1843.58	kPa	Joback Method
rinpol	2150.00		NIST Webbook
rinpol	2150.00		NIST Webbook
tb	786.62	K	Joback Method
tc	990.81	K	Joback Method
tf	524.09	K	Joback Method
vc	0.852	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	645.26	J/molxK	786.62	Joback Method
cpg	703.96	J/molxK	956.78	Joback Method
cpg	694.34	J/molxK	922.75	Joback Method
cpg	683.63	J/molxK	888.72	Joback Method
cpg	671.85	J/molxK	854.68	Joback Method
cpg	659.05	J/molxK	820.65	Joback Method
cpg	712.45	J/molxK	990.81	Joback Method
dvisc	0.0000631	Paxs	786.62	Joback Method

dvisc	0.0000763	Paxs	742.87	Joback Method
dvisc	0.0000944	Paxs	699.11	Joback Method
dvisc	0.0001202	Paxs	655.36	Joback Method
dvisc	0.0001585	Paxs	611.60	Joback Method
dvisc	0.0002181	Paxs	567.85	Joback Method
dvisc	0.0003164	Paxs	524.09	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R514233&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R514233&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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