

# Dichlorine monoxide

<b>Other names:</b>	Chlorine monooxide Chlorine monoxide Chlorine monoxide (Cl <sub>2</sub> O) Chlorine oxide Chlorine oxide (Cl <sub>2</sub> O) Cl <sub>2</sub> O Dichlorine oxide Dichloromonoxide Dichloroxide Hypochlorous anhydride
<b>Inchi:</b>	InChI=1S/Cl <sub>2</sub> O/c1-3-2
<b>InchiKey:</b>	RCJVRSBWZCINNQT-UHFFFAOYSA-N
<b>Formula:</b>	Cl <sub>2</sub> O
<b>SMILES:</b>	ClOCl
<b>Mol. weight [g/mol]:</b>	86.91
<b>CAS:</b>	7791-21-1

## Physical Properties

Property code	Value	Unit	Source
ea	1.30 ± 0.43	eV	NIST Webbook
gf	-179.74	kJ/mol	Joback Method
hf	-207.03	kJ/mol	Joback Method
hfus	5.34	kJ/mol	Joback Method
hvap	26.77	kJ/mol	Joback Method
ie	10.91 ± 0.02	eV	NIST Webbook
ie	11.20 ± 0.10	eV	NIST Webbook
ie	11.02	eV	NIST Webbook
ie	10.91 ± 0.02	eV	NIST Webbook
ie	10.94	eV	NIST Webbook
log10ws	-1.20		Crippen Method
logp	1.311		Crippen Method
mcvol	41.210	ml/mol	McGowan Method
pc	5836.07	kPa	Joback Method
tb	296.68	K	Joback Method
tc	479.25	K	Joback Method
tf	171.83	K	Joback Method
vc	0.151	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	47.60	J/molxK	296.68	Joback Method
cpg	47.57	J/molxK	327.11	Joback Method
cpg	47.62	J/molxK	357.54	Joback Method
cpg	47.75	J/molxK	387.97	Joback Method
cpg	47.95	J/molxK	418.39	Joback Method
cpg	48.21	J/molxK	448.82	Joback Method
cpg	48.53	J/molxK	479.25	Joback Method
dvisc	0.0019500	Paxs	171.83	Joback Method
dvisc	0.0011979	Paxs	192.64	Joback Method
dvisc	0.0008092	Paxs	213.45	Joback Method
dvisc	0.0005861	Paxs	234.25	Joback Method
dvisc	0.0004474	Paxs	255.06	Joback Method
dvisc	0.0003557	Paxs	275.87	Joback Method
dvisc	0.0002921	Paxs	296.68	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.57156e+01
Coeff. B	-2.98553e+03
Coeff. C	-6.32000e+00
Temperature range (K), min.	174.65
Temperature range (K), max.	275.35

## Sources

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C7791211&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307l>  
**Crippen Method:** [https://www.cheméo.com/doc/models/crippen\\_log10ws](https://www.cheméo.com/doc/models/crippen_log10ws)  
**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

**cpg:** Ideal gas heat capacity  
**dvisc:** Dynamic viscosity  
**ea:** Electron affinity  
**gf:** Standard Gibbs free energy of formation  
**hf:** Enthalpy of formation at standard conditions  
**hfus:** Enthalpy of fusion at standard conditions  
**hvap:** Enthalpy of vaporization at standard conditions  
**ie:** Ionization energy  
**log10ws:** Log10 of Water solubility in mol/l  
**logp:** Octanol/Water partition coefficient  
**mcvol:** McGowan's characteristic volume  
**pc:** Critical Pressure  
**pvap:** Vapor pressure  
**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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