

Hydroxylamine, O-methyl-

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| Other names: | CH3ONH2 Methoxyamine O-Methylhydroxylamine NCI-C60060 Hydroxylamine methyl ether Methoxyamine |
| Inchi: | InChI=1S/CH5NO/c1-3-2/h2H2,1H3 |
| InchiKey: | GMPKIPWJBDOURN-UHFFFAOYSA-N |
| Formula: | CH5NO |
| SMILES: | CON |
| Mol. weight [g/mol]: | 47.06 |
| CAS: | 67-62-9 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| affp | 844.80 | kJ/mol | NIST Webbook |
| basg | 812.30 | kJ/mol | NIST Webbook |
| gf | -81.01 | kJ/mol | Joback Method |
| hf | -162.40 | kJ/mol | Joback Method |
| hfus | 4.73 | kJ/mol | Joback Method |
| hvap | 30.87 | kJ/mol | Joback Method |
| ie | 10.28 | eV | NIST Webbook |
| ie | 9.55 | eV | NIST Webbook |
| log10ws | 0.25 | | Crippen Method |
| logp | -0.493 | | Crippen Method |
| mcvol | 40.800 | ml/mol | McGowan Method |
| pc | 6161.14 | kPa | Joback Method |
| tb | 317.23 | K | Joback Method |
| tc | 500.19 | K | Joback Method |
| tf | 206.52 | K | Joback Method |
| vc | 0.139 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-------|---------|-----------------|---------------|
| cpg | 80.29 | J/mol×K | 439.20 | Joback Method |
| cpg | 83.47 | J/mol×K | 469.70 | Joback Method |
| cpg | 67.25 | J/mol×K | 317.23 | Joback Method |
| cpg | 70.54 | J/mol×K | 347.72 | Joback Method |
| cpg | 73.81 | J/mol×K | 378.22 | Joback Method |
| cpg | 77.07 | J/mol×K | 408.71 | Joback Method |
| cpg | 86.61 | J/mol×K | 500.19 | Joback Method |
| hvapt | 36.90 | kJ/mol | 275.00 | NIST Webbook |
| hvapt | 38.00 | kJ/mol | 265.50 | NIST Webbook |

Sources

| | |
|------------------------|---|
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C67629&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |

Legend

| | |
|-----------------|---|
| affp: | Proton affinity |
| basg: | Gas basicity |
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| hvapt: | Enthalpy of vaporization at a given temperature |
| ie: | Ionization energy |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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