

# 2-Butyne-1,4-diol, diacetate

<b>Other names:</b>	Butynediol diacetate 1,4-Diacetoxy-2-butyne 2-Butynediol-1,4-diacetate but-2-yne-1,4-diyl diacetate
<b>Inchi:</b>	InChI=1S/C8H10O4/c1-7(9)11-5-3-4-6-12-8(2)10/h5-6H2,1-2H3
<b>InchiKey:</b>	TVIMIQVBDDNCCR-UHFFFAOYSA-N
<b>Formula:</b>	C8H10O4
<b>SMILES:</b>	CC(=O)OCC#CCOC(C)=O
<b>Mol. weight [g/mol]:</b>	170.16
<b>CAS:</b>	1573-17-7

## Physical Properties

Property code	Value	Unit	Source
gf	-248.56	kJ/mol	Joback Method
hf	-425.75	kJ/mol	Joback Method
hfus	25.17	kJ/mol	Joback Method
hvap	53.87	kJ/mol	Joback Method
log10ws	-0.69		Crippen Method
logp	0.116		Crippen Method
mcvol	129.860	ml/mol	McGowan Method
pc	3325.84	kPa	Joback Method
tb	544.02	K	Joback Method
tc	750.31	K	Joback Method
tf	430.34	K	Joback Method
vc	0.493	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	285.58	J/molxK	544.02	Joback Method
cpg	296.01	J/molxK	578.40	Joback Method
cpg	306.03	J/molxK	612.78	Joback Method
cpg	315.63	J/molxK	647.17	Joback Method
cpg	324.79	J/molxK	681.55	Joback Method

cpg	333.49	J/mol×K	715.93	Joback Method
cpg	341.73	J/mol×K	750.31	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	395.70	K	1.30	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1573177&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1573177&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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