

# 4-Octene

<b>Other names:</b>	Oct-4-ene
<b>Inchi:</b>	InChI=1S/C8H16/c1-3-5-7-8-6-4-2/h7-8H,3-6H2,1-2H3
<b>InchiKey:</b>	IRUCBBFNLDIMIK-UHFFFAOYSA-N
<b>Formula:</b>	C8H16
<b>SMILES:</b>	CCCC=CCCC
<b>Mol. weight [g/mol]:</b>	112.21
<b>CAS:</b>	592-99-4

## Physical Properties

Property code	Value	Unit	Source
gf	96.70	kJ/mol	Joback Method
hf	-91.23	kJ/mol	Joback Method
hfus	16.68	kJ/mol	Joback Method
hvap	33.36	kJ/mol	Joback Method
log10ws	-3.02		Crippen Method
logp	3.143		Crippen Method
mcvol	119.280	ml/mol	McGowan Method
pc	2673.54	kPa	Joback Method
rinpol	797.00		NIST Webbook
rinpol	797.00		NIST Webbook
rinpol	806.00		NIST Webbook
rinpol	806.00		NIST Webbook
rinpol	786.00		NIST Webbook
rinpol	806.00		NIST Webbook
rinpol	806.00		NIST Webbook
tb	395.25 ± 0.60	K	NIST Webbook
tb	393.85 ± 0.50	K	NIST Webbook
tb	395.45 ± 1.00	K	NIST Webbook
tc	558.45	K	Joback Method
tf	174.84	K	Joback Method
vc	0.464	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	218.05	J/mol×K	386.60	Joback Method
cpg	230.94	J/mol×K	415.24	Joback Method
cpg	243.29	J/mol×K	443.88	Joback Method
cpg	255.11	J/mol×K	472.53	Joback Method
cpg	266.43	J/mol×K	501.17	Joback Method
cpg	277.26	J/mol×K	529.81	Joback Method
cpg	287.62	J/mol×K	558.45	Joback Method
dvisc	0.0052396	Paxs	174.84	Joback Method
dvisc	0.0019294	Paxs	210.13	Joback Method
dvisc	0.0009470	Paxs	245.43	Joback Method
dvisc	0.0005559	Paxs	280.72	Joback Method
dvisc	0.0003675	Paxs	316.01	Joback Method
dvisc	0.0002641	Paxs	351.31	Joback Method
dvisc	0.0002015	Paxs	386.60	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52377e+01
Coeff. B	-3.64941e+03
Coeff. C	-5.17940e+01
Temperature range (K), min.	295.90
Temperature range (K), max.	419.45

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C592994&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C592994&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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