

GA29-2-O-«beta»-D-glucopyranoside, permethyl

Inchi:	InChI=1S/C31H46O11/c1-16-11-29-15-30(16,39-8)10-9-19(29)31-13-17(12-28(2,27(33)4
InchiKey:	AEIQEDJKRWZSNO-KGEVVGJPYSA-N
Formula:	C31H46O11
SMILES:	C=C1CC23CC1(OC)CCC2C12CC(OC4OC(COC)C(OC)C(OC)C4OC)CC(C)(C(=O)O1)C
Mol. weight [g/mol]:	594.69

Physical Properties

Property code	Value	Unit	Source
gf	-667.46	kJ/mol	Joback Method
hf	-1714.17	kJ/mol	Joback Method
hfus	60.88	kJ/mol	Joback Method
hvap	115.21	kJ/mol	Joback Method
log10ws	-3.54		Crippen Method
logp	2.434		Crippen Method
mcvol	434.160	ml/mol	McGowan Method
pc	885.77	kPa	Joback Method
rinpol	3592.00		NIST Webbook
rinpol	3591.00		NIST Webbook
tb	1270.43	K	Joback Method
tc	1560.31	K	Joback Method
tf	935.67	K	Joback Method
vc	1.623	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2057.50	J/mol×K	1270.43	Joback Method
cpg	2137.72	J/mol×K	1318.74	Joback Method
cpg	2225.40	J/mol×K	1367.06	Joback Method
cpg	2321.37	J/mol×K	1415.37	Joback Method
cpg	2426.48	J/mol×K	1463.68	Joback Method
cpg	2541.58	J/mol×K	1512.00	Joback Method
cpg	2667.51	J/mol×K	1560.31	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R178966&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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