

# 1-Methyl-4-(1-methylethyl)-3-(chloromethyl)benzene

<b>Inchi:</b>	InChI=1S/C11H15Cl/c1-8(2)11-5-4-9(3)6-10(11)7-12/h4-6,8H,7H2,1-3H3
<b>InchiKey:</b>	GUIBAZZCJFATIIY-UHFFFAOYSA-N
<b>Formula:</b>	C11H15Cl
<b>SMILES:</b>	Cc1ccc(C(C)C)c(CCl)c1
<b>Mol. weight [g/mol]:</b>	182.69

## Physical Properties

Property code	Value	Unit	Source
gf	120.52	kJ/mol	Joback Method
hf	-77.80	kJ/mol	Joback Method
hfus	18.18	kJ/mol	Joback Method
hvap	47.68	kJ/mol	Joback Method
log10ws	-4.16		Crippen Method
logp	3.857		Crippen Method
mcvol	154.330	ml/mol	McGowan Method
pc	2495.01	kPa	Joback Method
rinpol	1328.00		NIST Webbook
tb	524.71	K	Joback Method
tc	739.09	K	Joback Method
tf	280.11	K	Joback Method
vc	0.587	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	330.60	J/molxK	524.71	Joback Method
cpg	345.53	J/molxK	560.44	Joback Method
cpg	359.65	J/molxK	596.17	Joback Method
cpg	372.97	J/molxK	631.90	Joback Method
cpg	385.54	J/molxK	667.63	Joback Method
cpg	397.37	J/molxK	703.36	Joback Method
cpg	408.49	J/molxK	739.09	Joback Method
dvisc	0.0023785	Paxs	280.11	Joback Method
dvisc	0.0012121	Paxs	320.88	Joback Method

dvisc	0.0007191	Paxs	361.64	Joback Method
dvisc	0.0004742	Paxs	402.41	Joback Method
dvisc	0.0003376	Paxs	443.18	Joback Method
dvisc	0.0002545	Paxs	483.94	Joback Method
dvisc	0.0002005	Paxs	524.71	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R520411&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R520411&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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