

2,2,3,3,3-Pentafluoro-1-propanol

Other names:	1,1-Dihydroperfluoropropanol 1,1H-perfluoropropanol 1-Propanol, 2,2,3,3,3-pentafluoro- 1H,1H-Pentafluoropropanol 1H,1H-Pentafluoropropanol-1 2,2,3,3,3-Pentafluoropropanol C2F5CH2OH NSC 66413
Inchi:	InChI=1S/C3H3F5O/c4-2(5,1-9)3(6,7)8/h9H,1H2
InchiKey:	PSQZJKGXDGNDFP-UHFFFAOYSA-N
Formula:	C3H3F5O
SMILES:	OCC(F)(F)C(F)(F)F
Mol. weight [g/mol]:	150.05
CAS:	422-05-9

Physical Properties

Property code	Value	Unit	Source
chl	-1149.70 ± 2.80	kJ/mol	NIST Webbook
gf	-1130.81	kJ/mol	Joback Method
hf	-1255.53	kJ/mol	Joback Method
hfl	-1354.70 ± 2.90	kJ/mol	NIST Webbook
hfus	8.19	kJ/mol	Joback Method
hvap	44.40	kJ/mol	NIST Webbook
hvap	44.38	kJ/mol	NIST Webbook
hvap	41.30	kJ/mol	NIST Webbook
ie	11.68	eV	NIST Webbook
ie	11.20	eV	NIST Webbook
log10ws	-1.32		Crippen Method
logp	1.176		Crippen Method
mcvol	67.850	ml/mol	McGowan Method
pc	3901.37	kPa	Joback Method
tb	353.70	K	NIST Webbook
tb	355.00	K	NIST Webbook
tc	490.34	K	Joback Method
tf	192.18	K	Joback Method
vc	0.290	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	144.58	J/mol×K	350.11	Joback Method
cpg	151.21	J/mol×K	373.48	Joback Method
cpg	157.44	J/mol×K	396.85	Joback Method
cpg	163.28	J/mol×K	420.22	Joback Method
cpg	168.76	J/mol×K	443.59	Joback Method
cpg	173.89	J/mol×K	466.96	Joback Method
cpg	178.69	J/mol×K	490.34	Joback Method
hvapt	45.10	kJ/mol	313.01	Vapor pressure and liquid density of fluorinated alcohols: experimental, simulation and GC-SAFT-VR predictions
hvapt	47.00	kJ/mol	285.00	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	353.20	K	99.70	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.35801e+01
Coeff. B	-2.22287e+03
Coeff. C	-1.06110e+02
Temperature range (K), min.	273.00
Temperature range (K), max.	374.94

Sources

Vapor pressure and liquid density of fluorinated alcohols: experimental, Joback Method, PC-SAFT-VR predictions, McGowan Method:	https://www.doi.org/10.1016/j.fluid.2016.06.011
NIST Webbook:	https://en.wikipedia.org/wiki/Joback_method
The Yaws Handbook of Vapor Pressure: Crippen Method:	http://link.springer.com/article/10.1007/BF02311772
Crippen Method:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C422059&Units=SI
Fluorinated surfactants in solution: Diffusion coefficients of fluorinated alcohols in water:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
	http://pubs.acs.org/doi/abs/10.1021/ci9903071
	https://www.chemeo.com/doc/models/crippen_log10ws
	https://www.doi.org/10.1016/j.fluid.2015.06.006

Legend

chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfl:	Liquid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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