

# 1,4-Dibenzyloxybenzene

<b>Other names:</b>	Hydroquinone dibenzyl ether Benzene, 1,4-bis(phenylmethoxy)-
<b>Inchi:</b>	InChI=1S/C20H18O2/c1-3-7-17(8-4-1)15-21-19-11-13-20(14-12-19)22-16-18-9-5-2-6-10
<b>InchiKey:</b>	DYULYMCXVSRUPB-UHFFFAOYSA-N
<b>Formula:</b>	C20H18O2
<b>SMILES:</b>	c1ccc(COc2ccc(OCc3ccccc3)cc2)cc1
<b>Mol. weight [g/mol]:</b>	290.36
<b>CAS:</b>	621-91-0

## Physical Properties

Property code	Value	Unit	Source
gf	235.12	kJ/mol	Joback Method
hf	-22.45	kJ/mol	Joback Method
hfus	31.67	kJ/mol	Joback Method
hvap	72.42	kJ/mol	Joback Method
log10ws	-5.93		Crippen Method
logp	4.845		Crippen Method
mvol	233.120	ml/mol	McGowan Method
pc	2077.43	kPa	Joback Method
tb	786.86	K	Joback Method
tc	1035.84	K	Joback Method
tf	451.40	K	Joback Method
vc	0.868	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	658.13	J/molxK	786.86	Joback Method
cpg	728.10	J/molxK	994.34	Joback Method
cpg	716.92	J/molxK	952.84	Joback Method
cpg	704.40	J/molxK	911.35	Joback Method
cpg	690.47	J/molxK	869.85	Joback Method
cpg	675.08	J/molxK	828.36	Joback Method
cpg	738.04	J/molxK	1035.84	Joback Method

dvisc	0.0000600	Paxs	786.86	Joback Method
dvisc	0.0000765	Paxs	730.95	Joback Method
dvisc	0.0001014	Paxs	675.04	Joback Method
dvisc	0.0001414	Paxs	619.13	Joback Method
dvisc	0.0002106	Paxs	563.22	Joback Method
dvisc	0.0003426	Paxs	507.31	Joback Method
dvisc	0.0006288	Paxs	451.40	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C621910&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C621910&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/51-900-5/1-4-Dibenzyloxybenzene.pdf>

Generated by Cheméo on 2024-04-19 14:29:19.050042775 +0000 UTC m=+15826207.970620087.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.