

2,2-Diphenyl-4-mercaptobutyric acid

Inchi:	InChI=1S/C16H16O2S/c17-15(18)16(11-12-19,13-7-3-1-4-8-13)14-9-5-2-6-10-14/h1-10,1
InchiKey:	ZWILEONEUTXNPU-UHFFFAOYSA-N
Formula:	C16H16O2S
SMILES:	O=C(O)C(CCS)(c1ccccc1)c1ccccc1
Mol. weight [g/mol]:	272.36
CAS:	73758-57-3

Physical Properties

Property code	Value	Unit	Source
gf	75.15	kJ/mol	Joback Method
hf	-135.59	kJ/mol	Joback Method
hfus	27.59	kJ/mol	Joback Method
hvap	84.63	kJ/mol	Joback Method
log10ws	-3.67		Crippen Method
logp	3.377		Crippen Method
mcvol	212.570	ml/mol	McGowan Method
pc	2909.25	kPa	Joback Method
tb	824.52	K	Joback Method
tc	1068.34	K	Joback Method
tf	472.55	K	Joback Method
vc	0.783	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	596.03	J/molxK	824.52	Joback Method
cpg	608.30	J/molxK	865.16	Joback Method
cpg	619.53	J/molxK	905.79	Joback Method
cpg	629.85	J/molxK	946.43	Joback Method
cpg	639.38	J/molxK	987.07	Joback Method
cpg	648.24	J/molxK	1027.70	Joback Method
cpg	656.56	J/molxK	1068.34	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C73758573&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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