

# Octadecane, 1-iodo-

<b>Other names:</b>	1-Iodooctadecane Octadecyl iodide Stearyl iodide n-Octadecyl iodide
<b>Inchi:</b>	InChI=1S/C18H37I/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19/h2-18H2,1H3
<b>InchiKey:</b>	ZNJOCVLVYVOUGB-UHFFFAOYSA-N
<b>Formula:</b>	C18H37I
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCI
<b>Mol. weight [g/mol]:</b>	380.39
<b>CAS:</b>	629-93-6

## Physical Properties

Property code	Value	Unit	Source
gf	158.80	kJ/mol	Joback Method
hf	-337.98	kJ/mol	Joback Method
hfus	46.78	kJ/mol	Joback Method
hvap	109.30	kJ/mol	NIST Webbook
log10ws	-8.31		Crippen Method
logp	7.683		Crippen Method
mcvol	290.300	ml/mol	McGowan Method
pc	1135.20	kPa	Joback Method
rinpol	2290.00		NIST Webbook
rinpol	2240.00		NIST Webbook
rinpol	2290.00		NIST Webbook
rinpol	2272.00		NIST Webbook
ripol	2576.00		NIST Webbook
tb	656.20	K	NIST Webbook
tc	883.65	K	Joback Method
tf	307.00 ± 3.00	K	NIST Webbook
vc	1.131	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	905.12	J/mol×K	883.65	Joback Method
cpg	890.14	J/mol×K	853.77	Joback Method
cpg	874.42	J/mol×K	823.90	Joback Method
cpg	857.92	J/mol×K	794.02	Joback Method
cpg	840.60	J/mol×K	764.14	Joback Method
cpg	822.42	J/mol×K	734.26	Joback Method
cpg	803.35	J/mol×K	704.38	Joback Method
dvisc	0.0028659	Paxs	350.68	Joback Method
dvisc	0.0000975	Paxs	704.38	Joback Method
dvisc	0.0001325	Paxs	645.43	Joback Method
dvisc	0.0001913	Paxs	586.48	Joback Method
dvisc	0.0003000	Paxs	527.53	Joback Method
dvisc	0.0005268	Paxs	468.58	Joback Method
dvisc	0.0010877	Paxs	409.63	Joback Method
hvapt	77.20	kJ/mol	584.50	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	468.70	K	0.30	NIST Webbook
tbrp	468.50 ± 1.50	K	0.30	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.63563e+01
Coeff. B	-6.04763e+03
Coeff. C	-1.23560e+02
Temperature range (K), min.	499.92
Temperature range (K), max.	671.11

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C629936&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C629936&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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