

GLYCEROL TRICAPRYLATE

Other names:	1,2,3-propanetriol trioctanoate
	Caprylic acid triglyceride
	Caprylic triglyceride
	Caprylin
	Captex 8000
	Emalex KTG
	Glycerin tricaprylate
	Glycerol trioctanoate
	Maceight
	Miglyol 808
	NSC 4059
	Octanoic acid triglyceride
	Octanoic acid, 1,1',1''-(1,2,3-propanetriyl) ester
	Octanoic acid, 1,2,3-propanetriyl ester
	Octanoin, tri-
	Panacete 800
	RATO
	Sefsol 810
	Tricaprylic Glyceride
	Tricapryloylglycerol
	Tricaprylyl glycerin
	Trioctanoylglycerol
	glyceryl tricaprylate
	glyceryl trioctanoate
	tricaprilin
	tricaprylin
	trioctanoin
Inchi:	InChI=1S/C27H50O6/c1-4-7-10-13-16-19-25(28)31-22-24(33-27(30)21-18-15-12-9-6-3)2
InchiKey:	VLPFTAMPNXLGLX-UHFFFAOYSA-N
Formula:	C27H50O6
SMILES:	CCCCCCCC(=O)OCC(COC(=O)CCCCCCC)OC(=O)CCCCCCC
Mol. weight [g/mol]:	470.69

Physical Properties

Property code	Value	Unit	Source
chs	-15259.00 ± 1.80	kJ/mol	NIST Webbook

hf	-1516.05	kJ/mol	Cheméo Relay (1.0)
hfus	70.52	kJ/mol	Joback Method
hvap	135.40 ± 4.70	kJ/mol	NIST Webbook
ie	9.61	eV	Cheméo Relay (1.0)
log10ws	-6.52		Cheméo Relay (1.0)
logp	7.066		Crippen Method
mcvol	413.610	ml/mol	McGowan Method
pc	744.88	kPa	Joback Method
rinpol	2958.00		NIST Webbook
tb	670.51	K	Cheméo Relay (1.0)
tc	836.00	K	Critical properties, heat capacities, and thermal diffusivities of four saturated triglycerides
tf	294.95	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1458.42	J/mol×K	1045.59	Joback Method
cpg	1477.45	J/mol×K	1088.41	Joback Method
cpg	1494.01	J/mol×K	1131.24	Joback Method
cpg	1508.15	J/mol×K	1174.06	Joback Method
cpg	1519.94	J/mol×K	1216.88	Joback Method
cpg	1529.42	J/mol×K	1259.70	Joback Method
cpg	1536.66	J/mol×K	1302.53	Joback Method
cpl	920.00	J/mol×K	338.00	NIST Webbook
dvisc	0.0000964	Pa×s	670.54	Joback Method
dvisc	0.0000529	Pa×s	745.55	Joback Method
dvisc	0.0000324	Pa×s	820.56	Joback Method
dvisc	0.0000216	Pa×s	895.57	Joback Method
dvisc	0.0000153	Pa×s	970.58	Joback Method
dvisc	0.0000114	Pa×s	1045.59	Joback Method
dvisc	0.0002041	Pa×s	595.53	Joback Method
hvapt	118.70	kJ/mol	386.00	NIST Webbook
hvapt	116.00	kJ/mol	424.50	NIST Webbook

Datasets

Mass density, kg/m3

Temperature, K - Liquid	Pressure, kPa - Liquid	Mass density, kg/m3 - Liquid
293.15	102.00	952.7
298.15	102.00	948.9
303.15	102.00	944.9
313.15	102.00	937.4
333.15	102.00	922.5
353.15	102.00	907.7
298.15	120.00	949.9
298.15	9830.00	955.5
298.15	20100.00	961.3
298.15	30020.00	966.6
298.15	40050.00	971.9
298.15	50040.00	977.1
Reference		https://www.doi.org/10.1021/acs.jced.8b01046

Sources

Cheméo Relay (1.0):
Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws
Physical properties of systems of interest to the edible oil industry: The models and densities of model systems formed using PG-SAGT, Viscosity of Fatty Mixtures: Experimental Data and Prediction: Joback Method: <https://www.doi.org/10.1016/j.jct.2017.06.012>
Critical properties, heat capacities, and thermal diffusivities of four saturated triglycerides: <https://www.doi.org/10.1021/acs.jced.8b01046>
Cheméo Relay (1.0, beta): <https://www.doi.org/10.1021/je700293s>
McGowan Method: https://en.wikipedia.org/wiki/Joback_method
Crippen Method: <https://www.doi.org/10.1016/j.jct.2017.07.006>
<http://webbook.nist.gov/cgi/cbook.cgi?ID=C538238&Units=SI>
<http://link.springer.com/article/10.1007/BF02311772>
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Legend

- chs: Standard solid enthalpy of combustion
- cpg: Ideal gas heat capacity

cpl:	Liquid phase heat capacity
dvisc:	Dynamic viscosity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rho:	Liquid Density
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point

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