

# Bromomethyl radical

**Inchi:** InChI=1S/CH2Br/c1-2/h1H2  
**InchiKey:** AOJDZKCUAATBGE-UHFFFAOYSA-N  
**Formula:** CH2Br  
**SMILES:** [CH2]Br  
**Mol. weight [g/mol]:** 93.93  
**CAS:** 16519-97-4

## Physical Properties

Property code	Value	Unit	Source
ea	1.94	eV	NIST Webbook
ea	0.97 ± 0.16	eV	NIST Webbook
ea	0.79 ± 0.14	eV	NIST Webbook
gf	24.24	kJ/mol	Joback Method
hf	18.17	kJ/mol	Joback Method
hfus	5.31	kJ/mol	Joback Method
hvap	24.11	kJ/mol	Joback Method
ie	8.61 ± 0.01	eV	NIST Webbook
ie	8.72 ± 0.01	eV	NIST Webbook
ie	8.34 ± 0.11	eV	NIST Webbook
ie	8.61 ± 0.01	eV	NIST Webbook
ie	9.30	eV	NIST Webbook
log10ws	-0.78		Crippen Method
logp	1.173		Crippen Method
mcvol	40.300	ml/mol	McGowan Method
pc	6932.88	kPa	Joback Method
tb	287.74	K	Joback Method
tc	468.60	K	Joback Method
tf	177.20	K	Joback Method
vc	0.144	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	36.65	J/mol×K	287.74	Joback Method

cpg	39.43	J/mol×K	317.88	Joback Method
cpg	41.89	J/mol×K	348.03	Joback Method
cpg	44.07	J/mol×K	378.17	Joback Method
cpg	45.98	J/mol×K	408.32	Joback Method
cpg	47.64	J/mol×K	438.46	Joback Method
cpg	49.09	J/mol×K	468.60	Joback Method
dvisc	0.0005163	Paxs	177.20	Joback Method
dvisc	0.0004556	Paxs	195.62	Joback Method
dvisc	0.0004108	Paxs	214.05	Joback Method
dvisc	0.0003765	Paxs	232.47	Joback Method
dvisc	0.0003496	Paxs	250.89	Joback Method
dvisc	0.0003279	Paxs	269.32	Joback Method
dvisc	0.0003100	Paxs	287.74	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16519974&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16519974&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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