

# Benzenamine, 3,5-dinitro-

<b>Other names:</b>	Aniline, 3,5-dinitro- 3,5-Dinitroaniline Benzeneamine, 3,5-dinitro-
<b>Inchi:</b>	InChI=1S/C6H5N3O4/c7-4-1-5(8(10)11)3-6(2-4)9(12)13/h1-3H,7H2
<b>InchiKey:</b>	MPBZUKLDHPOCLS-UHFFFAOYSA-N
<b>Formula:</b>	C6H5N3O4
<b>SMILES:</b>	<chem>Nc1cc([N+](=O)[O-])cc([N+](=O)[O-])c1</chem>
<b>Mol. weight [g/mol]:</b>	183.12
<b>CAS:</b>	618-87-1

## Physical Properties

Property code	Value	Unit	Source
chs	-3037.00	kJ/mol	NIST Webbook
gf	230.34	kJ/mol	Joback Method
hf	58.69	kJ/mol	Joback Method
hfs	-39.00	kJ/mol	NIST Webbook
hfus	32.48	kJ/mol	Joback Method
hvap	76.37	kJ/mol	Joback Method
log10ws	-2.41		Crippen Method
logp	1.085		Crippen Method
mcvol	116.460	ml/mol	McGowan Method
pc	5015.69	kPa	Joback Method
rinpol	321.42		NIST Webbook
tb	749.53	K	Joback Method
tc	1034.10	K	Joback Method
tf	579.32	K	Joback Method
vc	0.457	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	298.44	J/molxK	749.53	Joback Method
cpg	306.66	J/molxK	796.96	Joback Method
cpg	314.01	J/molxK	844.39	Joback Method

cpg	320.54	J/mol×K	891.81	Joback Method
cpg	326.32	J/mol×K	939.24	Joback Method
cpg	331.39	J/mol×K	986.67	Joback Method
cpg	335.82	J/mol×K	1034.10	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C618871&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C618871&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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