

# 3-Benzylidene-2,4-pentanedione

<b>Other names:</b>	2,4-Pentanedione, 3-(phenylmethylene)-Benzalacetylacetone Benzylidene acetylacetone 1,1-Diacetyl-2-phenylethylene 2-Acetyl-1-phenyl but-1-en-3-one 2,4-Pentanedione, 3-benzylidene-3-Benzylideneacetylacetone (2,2-Diacetylvinyl)benzene NSC 35140 3-(phenylmethylene)pentane-2,4-dione
<b>Inchi:</b>	InChI=1S/C12H12O2/c1-9(13)12(10(2)14)8-11-6-4-3-5-7-11/h3-8H,1-2H3
<b>InchiKey:</b>	NYRGMNMVISROGJ-UHFFFAOYSA-N
<b>Formula:</b>	C12H12O2
<b>SMILES:</b>	CC(=O)C(=Cc1ccccc1)C(C)=O
<b>Mol. weight [g/mol]:</b>	188.22
<b>CAS:</b>	4335-90-4

## Physical Properties

Property code	Value	Unit	Source
gf	-23.60	kJ/mol	Joback Method
hf	-172.21	kJ/mol	Joback Method
hfus	22.97	kJ/mol	Joback Method
hvap	58.11	kJ/mol	Joback Method
log10ws	-2.53		Crippen Method
logp	2.248		Crippen Method
mcvol	155.020	ml/mol	McGowan Method
pc	2902.98	kPa	Joback Method
tb	612.42	K	Joback Method
tc	841.51	K	Joback Method
tf	332.24	K	Joback Method
vc	0.593	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	366.36	J/mol×K	612.42	Joback Method
cpg	380.05	J/mol×K	650.60	Joback Method
cpg	392.76	J/mol×K	688.78	Joback Method
cpg	404.52	J/mol×K	726.96	Joback Method
cpg	415.42	J/mol×K	765.15	Joback Method
cpg	425.51	J/mol×K	803.33	Joback Method
cpg	434.86	J/mol×K	841.51	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4335904&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4335904&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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