

# cis-1-Methyl-3-isopropylcyclopentane

<b>Other names:</b>	1-cis-3-methylisopropylcyclopentane CIS-1-N-ISOPROPYL-3-METHYLCYCLOPENTANE Cyclopentane, 1-methyl-3-(1-methylethyl)-, cis
<b>Inchi:</b>	InChI=1S/C9H18/c1-7(2)9-5-4-8(3)6-9/h7-9H,4-6H2,1-3H3/t8-,9+/m1/s1
<b>InchiKey:</b>	CDTDMKCVKCGRPD-BDAKNGLRSA-N
<b>Formula:</b>	C9H18
<b>SMILES:</b>	CC1CCC(C(C)C)C1
<b>Mol. weight [g/mol]:</b>	126.24

## Physical Properties

Property code	Value	Unit	Source
gf	51.30	kJ/mol	Joback Method
hf	-194.23	kJ/mol	Joback Method
hfus	10.55	kJ/mol	Joback Method
hvap	35.19	kJ/mol	Joback Method
log10ws	-2.76		Crippen Method
logp	3.079		Crippen Method
mcvol	126.810	ml/mol	McGowan Method
pc	2693.00	kPa	Joback Method
rinpol	861.00		NIST Webbook
rinpol	869.00		NIST Webbook
rinpol	872.00		NIST Webbook
rinpol	866.00		NIST Webbook
rinpol	856.00		NIST Webbook
rinpol	869.00		NIST Webbook
rinpol	865.00		NIST Webbook
rinpol	861.00		NIST Webbook
rinpol	859.70		NIST Webbook
rinpol	856.00		NIST Webbook
tb	415.49	K	Joback Method
tc	610.70	K	Joback Method
tf	182.85	K	Joback Method
vc	0.473	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	255.75	J/molxK	415.49	Joback Method
cpg	339.06	J/molxK	578.17	Joback Method
cpg	323.99	J/molxK	545.63	Joback Method
cpg	308.15	J/molxK	513.10	Joback Method
cpg	291.51	J/molxK	480.56	Joback Method
cpg	274.05	J/molxK	448.03	Joback Method
cpg	353.37	J/molxK	610.70	Joback Method
dvisc	0.0002863	Paxs	415.49	Joback Method
dvisc	0.0003541	Paxs	376.72	Joback Method
dvisc	0.0004600	Paxs	337.94	Joback Method
dvisc	0.0006395	Paxs	299.17	Joback Method
dvisc	0.0009806	Paxs	260.40	Joback Method
dvisc	0.0017463	Paxs	221.62	Joback Method
dvisc	0.0039722	Paxs	182.85	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=516">https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=516</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R92997&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R92997&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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