

Methyl (E)-trans-«alpha»-bergamota-2,10-dien-12-oate

Inchi: InChI=1S/C16H24O2/c1-11-7-8-13-10-14(11)16(13,3)9-5-6-12(2)15(17)18-4/h6-7,13-14H
InchiKey: ZTBCGQOFWCYELA-SUCCEKNJSA-N
Formula: C16H24O2
SMILES: COC(=O)C(C)=CCCC1(C)C2CC=C(C)C1C2
Mol. weight [g/mol]: 248.36

Physical Properties

Property code	Value	Unit	Source
gf	38.12	kJ/mol	Joback Method
hf	-330.29	kJ/mol	Joback Method
hfus	28.65	kJ/mol	Joback Method
hvap	59.90	kJ/mol	Joback Method
log10ws	-4.15		Crippen Method
logp	3.878		Crippen Method
mcvol	213.420	ml/mol	McGowan Method
pc	1837.26	kPa	Joback Method
rinpol	1785.00		NIST Webbook
tb	663.27	K	Joback Method
tc	872.43	K	Joback Method
tf	388.50	K	Joback Method
vc	0.826	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	604.93	J/mol×K	663.27	Joback Method
cpg	623.53	J/mol×K	698.13	Joback Method
cpg	641.22	J/mol×K	732.99	Joback Method
cpg	658.16	J/mol×K	767.85	Joback Method
cpg	674.47	J/mol×K	802.71	Joback Method
cpg	690.32	J/mol×K	837.57	Joback Method
cpg	705.85	J/mol×K	872.43	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R233711&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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