

# 4'-Methylbutyrophenone

<b>Other names:</b>	4-Methylbutyrophenone p-Methyl butyrophenone 1-Butanone, 1-(4-methylphenyl)- Butyrophenone, 4'-methyl- 1-(4-Tolyl)-1-butanone
<b>Inchi:</b>	InChI=1S/C11H14O/c1-3-4-11(12)10-7-5-9(2)6-8-10/h5-8H,3-4H2,1-2H3
<b>InchiKey:</b>	CIYAESDXUTVTAL-UHFFFAOYSA-N
<b>Formula:</b>	C11H14O
<b>SMILES:</b>	CCCC(=O)c1ccc(C)cc1
<b>Mol. weight [g/mol]:</b>	162.23
<b>CAS:</b>	4160-52-5

## Physical Properties

Property code	Value	Unit	Source
gf	15.60	kJ/mol	Joback Method
hf	-157.89	kJ/mol	Joback Method
hfus	19.50	kJ/mol	Joback Method
hvap	49.76	kJ/mol	Joback Method
ie	8.80 ± 0.20	eV	NIST Webbook
log10ws	-3.44		Crippen Method
logp	2.978		Crippen Method
mcvol	143.660	ml/mol	McGowan Method
pc	2784.72	kPa	Joback Method
tb	536.61	K	Joback Method
tc	749.19	K	Joback Method
tf	302.60	K	Joback Method
vc	0.549	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	321.61	J/mol×K	536.61	Joback Method
cpg	336.08	J/mol×K	572.04	Joback Method
cpg	349.73	J/mol×K	607.47	Joback Method

cpg	362.58	J/mol×K	642.90	Joback Method
cpg	374.66	J/mol×K	678.33	Joback Method
cpg	386.00	J/mol×K	713.76	Joback Method
cpg	396.64	J/mol×K	749.19	Joback Method
dvisc	0.0022892	Paxs	302.60	Joback Method
dvisc	0.0012638	Paxs	341.60	Joback Method
dvisc	0.0007880	Paxs	380.60	Joback Method
dvisc	0.0005365	Paxs	419.60	Joback Method
dvisc	0.0003899	Paxs	458.61	Joback Method
dvisc	0.0002979	Paxs	497.61	Joback Method
dvisc	0.0002367	Paxs	536.61	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4160525&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4160525&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/52-224-5/4-Methylbutyrophenone.pdf>

Generated by Cheméo on 2024-04-23 06:25:24.885212981 +0000 UTC m=+16142773.805790293.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.