

# Bicyclo[3.1.1]heptane-2-methanol, 6,6-dimethyl-

Other names:	6,6-Dimethyl-bicyclo[3.1.1]heptane-2-methanol Myrtanol (1«alpha»,2«alpha»,5«alpha»)-6,6-dimethylbicyclo[3.1.1]heptane-2-methanol
Inchi:	InChI=1S/C10H18O/c1-10(2)8-4-3-7(6-11)9(10)5-8/h7-9,11H,3-6H2,1-2H3
InchiKey:	LDWAIHWGMRVEFR-UHFFFAOYSA-N
Formula:	C10H18O
SMILES:	CC1(C)C2CCC(CO)C1C2
Mol. weight [g/mol]:	154.25
CAS:	514-99-8

## Physical Properties

Property code	Value	Unit	Source
gf	-15.01	kJ/mol	Joback Method
hf	-287.96	kJ/mol	Joback Method
hfus	15.76	kJ/mol	Joback Method
hvap	52.76	kJ/mol	Joback Method
log10ws	-2.09		Crippen Method
logp	2.051		Crippen Method
mcvol	135.910	ml/mol	McGowan Method
pc	3012.33	kPa	Joback Method
ripol	1256.00		NIST Webbook
ripol	1240.00		NIST Webbook
ripol	1249.00		NIST Webbook
ripol	1257.10		NIST Webbook
ripol	1250.00		NIST Webbook
ripol	1249.00		NIST Webbook
ripol	1262.00		NIST Webbook
ripol	1262.40		NIST Webbook
ripol	1258.00		NIST Webbook
ripol	1876.00		NIST Webbook
ripol	1889.00		NIST Webbook
ripol	1847.00		NIST Webbook
ripol	1889.00		NIST Webbook
ripol	1889.00		NIST Webbook
ripol	1876.00		NIST Webbook
tb	529.03	K	Joback Method
tc	722.31	K	Joback Method

tf	311.06	K	Joback Method
vc	0.516	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	354.68	J/mol×K	529.03	Joback Method
cpg	370.78	J/mol×K	561.24	Joback Method
cpg	385.90	J/mol×K	593.46	Joback Method
cpg	400.13	J/mol×K	625.67	Joback Method
cpg	413.58	J/mol×K	657.88	Joback Method
cpg	426.35	J/mol×K	690.10	Joback Method
cpg	438.56	J/mol×K	722.31	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15358915&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15358915&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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