

# Phenol, 3-ethyl-5-methyl-

<b>Other names:</b>	1-Hydroxy-3-methyl-5-ethylbenzene 3-Ethyl-5-methylphenol 3-Methyl-5-ethylphenol 5-Ethyl-3-methylphenol 5-ethyl-m-cresol m-Cresol, 5-ethyl-
<b>Inchi:</b>	InChI=1S/C9H12O/c1-3-8-4-7(2)5-9(10)6-8/h4-6,10H,3H2,1-2H3
<b>InchiKey:</b>	XTCHLXABLZQNNN-UHFFFAOYSA-N
<b>Formula:</b>	C9H12O
<b>SMILES:</b>	CCc1cc(C)cc(O)c1
<b>Mol. weight [g/mol]:</b>	136.19
<b>CAS:</b>	698-71-5

## Physical Properties

Property code	Value	Unit	Source
gf	-26.94	kJ/mol	Joback Method
hf	-181.34	kJ/mol	Joback Method
hfus	18.50	kJ/mol	Joback Method
hvap	51.58	kJ/mol	Joback Method
log10ws	-2.30		Crippen Method
logp	2.263		Crippen Method
mcvol	119.780	ml/mol	McGowan Method
pc	3824.55	kPa	Joback Method
ripol	1520.00		NIST Webbook
ripol	1520.00		NIST Webbook
ripol	1520.00		NIST Webbook
tb	517.60	K	Joback Method
tc	740.98	K	Joback Method
tf	341.85	K	Joback Method
vc	0.398	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
---------------	-------	------	-----------------	--------

cpg	267.07	J/molxK	517.60	Joback Method
cpg	279.62	J/molxK	554.83	Joback Method
cpg	291.33	J/molxK	592.06	Joback Method
cpg	302.26	J/molxK	629.29	Joback Method
cpg	312.50	J/molxK	666.52	Joback Method
cpg	322.10	J/molxK	703.75	Joback Method
cpg	331.15	J/molxK	740.98	Joback Method
dvisc	0.0011514	Paxs	371.14	Joback Method
dvisc	0.0026759	Paxs	341.85	Joback Method
dvisc	0.0005605	Paxs	400.43	Joback Method
dvisc	0.0003010	Paxs	429.73	Joback Method
dvisc	0.0001750	Paxs	459.02	Joback Method
dvisc	0.0001085	Paxs	488.31	Joback Method
dvisc	0.0000711	Paxs	517.60	Joback Method
hvapt	55.00	kJ/mol	494.50	NIST Webbook
hvapt	58.50	kJ/mol	445.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54470e+01
Coeff. B	-4.61795e+03
Coeff. C	-8.26940e+01
Temperature range (K), min.	387.32
Temperature range (K), max.	538.31

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C698715&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C698715&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/52-359-6/Phenol-3-ethyl-5-methyl.pdf>

Generated by Cheméo on 2024-04-23 14:22:13.19173396 +0000 UTC m=+16171382.112311292.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.