

# 2-(5-Methyl-5-vinyltetrahydrofuran-2-yl)propan-2-yl 3-chlorobenzoate

InChI: InChI=1S/C17H21ClO3/c1-5-17(4)10-9-14(20-17)16(2,3)21-15(19)12-7-6-8-13(18)11-12/1-17H1-17,19H2,20H2,21H3  
InChIKey: NWWVKPPRSHNJPP-UHFFFAOYSA-N

Formula: C17H21ClO3

SMILES: C=CC1(C)CCC(C(C)(C)OC(=O)c2cccc(Cl)c2)O1

Mol. weight [g/mol]: 308.80

## Physical Properties

Property code	Value	Unit	Source
gf	-22.90	kJ/mol	Joback Method
hf	-389.63	kJ/mol	Joback Method
hfus	28.41	kJ/mol	Joback Method
hvap	71.26	kJ/mol	Joback Method
log10ws	-5.34		Crippen Method
logp	4.399		Crippen Method
mcvol	237.020	ml/mol	McGowan Method
pc	1921.98	kPa	Joback Method
rinpol	2024.00		NIST Webbook
rinpol	2024.00		NIST Webbook
tb	764.99	K	Joback Method
tc	1005.35	K	Joback Method
tf	480.16	K	Joback Method
vc	0.881	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	683.15	J/mol×K	764.99	Joback Method
cpg	701.18	J/mol×K	805.05	Joback Method
cpg	718.36	J/mol×K	845.11	Joback Method
cpg	734.87	J/mol×K	885.17	Joback Method
cpg	750.92	J/mol×K	925.23	Joback Method
cpg	766.72	J/mol×K	965.29	Joback Method
cpg	782.47	J/mol×K	1005.35	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373576&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373576&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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