

# Heptanoic acid, 2-phenylethyl ester

<b>Other names:</b>	Phenylethyl heptanoate Phenylethyl heptylate Phenylethyl n-heptanoate 2-phenylethyl heptanoate
<b>Inchi:</b>	InChI=1S/C15H22O2/c1-2-3-4-8-11-15(16)17-13-12-14-9-6-5-7-10-14/h5-7,9-10H,2-4,8,1
<b>InchiKey:</b>	YMPDQHXHLKFWNN-UHFFFAOYSA-N
<b>Formula:</b>	C15H22O2
<b>SMILES:</b>	CCCCCCC(=O)OCCc1ccccc1
<b>Mol. weight [g/mol]:</b>	234.33
<b>CAS:</b>	5454-11-5

## Physical Properties

Property code	Value	Unit	Source
gf	-46.09	kJ/mol	Joback Method
hf	-361.20	kJ/mol	Joback Method
hfus	31.43	kJ/mol	Joback Method
hvap	60.42	kJ/mol	Joback Method
log10ws	-4.07		Crippen Method
logp	3.743		Crippen Method
mcvol	205.890	ml/mol	McGowan Method
pc	1921.98	kPa	Joback Method
rinpol	1707.00		NIST Webbook
rinpol	1718.00		NIST Webbook
rinpol	1707.00		NIST Webbook
rinpol	1740.00		NIST Webbook
rinpol	1718.00		NIST Webbook
ripol	2233.00		NIST Webbook
tb	645.57	K	Joback Method
tc	843.08	K	Joback Method
tf	357.39	K	Joback Method
vc	0.791	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	546.39	J/mol×K	645.57	Joback Method
cpg	563.17	J/mol×K	678.49	Joback Method
cpg	579.02	J/mol×K	711.41	Joback Method
cpg	593.96	J/mol×K	744.32	Joback Method
cpg	608.02	J/mol×K	777.24	Joback Method
cpg	621.23	J/mol×K	810.16	Joback Method
cpg	633.62	J/mol×K	843.08	Joback Method
dvisc	0.0019370	Paxs	357.39	Joback Method
dvisc	0.0009578	Paxs	405.42	Joback Method
dvisc	0.0005498	Paxs	453.45	Joback Method
dvisc	0.0003510	Paxs	501.48	Joback Method
dvisc	0.0002424	Paxs	549.51	Joback Method
dvisc	0.0001776	Paxs	597.54	Joback Method
dvisc	0.0001364	Paxs	645.57	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5454115&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5454115&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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