

# Benzene, 3-butyl-1(chloromethyl)-4-methyl

<b>Inchi:</b>	InChI=1S/C11H15Cl/c1-3-4-11-7-10(8-12)6-5-9(11)2/h5-7H,3-4,8H2,1-2H3
<b>InchiKey:</b>	HZVZMQXEFWZFHU-UHFFFAOYSA-N
<b>Formula:</b>	C11H15Cl
<b>SMILES:</b>	CCCC1cc(CCl)ccc1C
<b>Mol. weight [g/mol]:</b>	182.69

## Physical Properties

Property code	Value	Unit	Source
gf	122.96	kJ/mol	Joback Method
hf	-72.52	kJ/mol	Joback Method
hfus	21.71	kJ/mol	Joback Method
hvap	48.06	kJ/mol	Joback Method
log10ws	-4.20		Crippen Method
logp	3.686		Crippen Method
mvol	154.330	ml/mol	McGowan Method
pc	2475.19	kPa	Joback Method
rinpol	1452.00		NIST Webbook
rinpol	1452.00		NIST Webbook
tb	525.15	K	Joback Method
tc	734.82	K	Joback Method
tf	295.11	K	Joback Method
vc	0.593	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	330.32	J/molxK	525.15	Joback Method
cpg	344.85	J/molxK	560.10	Joback Method
cpg	358.61	J/molxK	595.04	Joback Method
cpg	371.63	J/molxK	629.99	Joback Method
cpg	383.92	J/molxK	664.93	Joback Method
cpg	395.52	J/molxK	699.88	Joback Method
cpg	406.46	J/molxK	734.82	Joback Method
dvisc	0.0018269	Paxs	295.11	Joback Method

dvisc	0.0010391	Paxs	333.45	Joback Method
dvisc	0.0006639	Paxs	371.79	Joback Method
dvisc	0.0004612	Paxs	410.13	Joback Method
dvisc	0.0003410	Paxs	448.47	Joback Method
dvisc	0.0002645	Paxs	486.81	Joback Method
dvisc	0.0002128	Paxs	525.15	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R132079&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R132079&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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