

# Diethylmalonic acid, monochloride, 2-naphthylmethyl ester

<b>Inchi:</b>	InChI=1S/C18H19ClO3/c1-3-18(4-2,16(19)20)17(21)22-12-13-9-10-14-7-5-6-8-15(14)11-
<b>InchiKey:</b>	IOYZUTNBQPUOLE-UHFFFAOYSA-N
<b>Formula:</b>	C18H19ClO3
<b>SMILES:</b>	CCC(CC)(C(=O)Cl)C(=O)OCc1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	318.80

## Physical Properties

Property code	Value	Unit	Source
gf	-61.82	kJ/mol	Joback Method
hf	-380.59	kJ/mol	Joback Method
hfus	34.22	kJ/mol	Joback Method
hvap	79.23	kJ/mol	Joback Method
log10ws	-5.63		Crippen Method
logp	4.455		Crippen Method
mcvol	242.510	ml/mol	McGowan Method
pc	1900.26	kPa	Joback Method
rinsol	2352.00		NIST Webbook
tb	826.24	K	Joback Method
tc	1056.94	K	Joback Method
tf	518.69	K	Joback Method
vc	0.925	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	687.93	J/molxK	826.24	Joback Method
cpg	701.50	J/molxK	864.69	Joback Method
cpg	714.06	J/molxK	903.14	Joback Method
cpg	725.71	J/molxK	941.59	Joback Method
cpg	736.55	J/molxK	980.04	Joback Method
cpg	746.69	J/molxK	1018.49	Joback Method
cpg	756.21	J/molxK	1056.94	Joback Method
dvisc	0.0009058	Paxs	518.69	Joback Method
dvisc	0.0005656	Paxs	569.95	Joback Method

dvisc	0.0003817	Paxs	621.21	Joback Method
dvisc	0.0002735	Paxs	672.47	Joback Method
dvisc	0.0002055	Paxs	723.72	Joback Method
dvisc	0.0001603	Paxs	774.98	Joback Method
dvisc	0.0001290	Paxs	826.24	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370560&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370560&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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