

# Benzoic acid, 4-(methylamino)-

<b>Other names:</b>	Benzoic acid, p-(methylamino)- p-(Methylamino)benzoic acid N-Methyl-4-aminobenzoic acid 4-(Methylamino)benzoic acid 4-(N-Methylamino)benzoic acid p-(N-Methylamino)benzoic acid
<b>Inchi:</b>	InChI=1S/C8H9NO2/c1-9-7-4-2-6(3-5-7)8(10)11/h2-5,9H,1H3,(H,10,11)
<b>InchiKey:</b>	ZVIDMSBTYRSMAR-UHFFFAOYSA-N
<b>Formula:</b>	C8H9NO2
<b>SMILES:</b>	CNc1ccc(C(=O)O)cc1
<b>Mol. weight [g/mol]:</b>	151.16
<b>CAS:</b>	10541-83-0

## Physical Properties

Property code	Value	Unit	Source
gf	-57.09	kJ/mol	Joback Method
hf	-194.73	kJ/mol	Joback Method
hfus	20.91	kJ/mol	Joback Method
hvap	66.20	kJ/mol	Joback Method
ie	8.10	eV	NIST Webbook
ie	7.30	eV	NIST Webbook
log10ws	-1.54		Crippen Method
logp	1.426		Crippen Method
mcvol	117.240	ml/mol	McGowan Method
pc	4432.62	kPa	Joback Method
tb	610.32	K	Joback Method
tc	817.72	K	Joback Method
tf	382.27	K	Joback Method
vc	0.435	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.76	J/molxK	610.32	Joback Method

cpg	288.23	J/mol×K	644.89	Joback Method
cpg	297.10	J/mol×K	679.45	Joback Method
cpg	305.39	J/mol×K	714.02	Joback Method
cpg	313.13	J/mol×K	748.59	Joback Method
cpg	320.34	J/mol×K	783.15	Joback Method
cpg	327.05	J/mol×K	817.72	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10541830&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10541830&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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