

# 3-n-Propyl-2,4-pentanedione

<b>Other names:</b>	2,4-Pentanedione, 3-propyl- 3-Propylacetylacetone 3-propyl-2,4-pentadione 3-propylpentane-2,4-dione
<b>Inchi:</b>	InChI=1S/C8H14O2/c1-4-5-8(6(2)9)7(3)10/h8H,4-5H2,1-3H3
<b>InchiKey:</b>	AQGSZYZZVTYOMQ-UHFFFAOYSA-N
<b>Formula:</b>	C8H14O2
<b>SMILES:</b>	CCCC(C(C)=O)C(C)=O
<b>Mol. weight [g/mol]:</b>	142.20
<b>CAS:</b>	1540-35-8

## Physical Properties

Property code	Value	Unit	Source
gf	-243.80	kJ/mol	Joback Method
hf	-453.10	kJ/mol	NIST Webbook
hfus	16.15	kJ/mol	Joback Method
hvap	46.51	kJ/mol	Joback Method
log10ws	-0.88		Aqueous Solubility Prediction Method
logp	1.581		Crippen Method
mcvol	126.720	ml/mol	McGowan Method
pc	2915.53	kPa	Joback Method
tb	489.74	K	Joback Method
tc	679.53	K	Joback Method
tf	264.78	K	Joback Method
vc	0.489	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	275.42	J/molxK	489.74	Joback Method
cpg	330.55	J/molxK	647.90	Joback Method
cpg	320.57	J/molxK	616.27	Joback Method
cpg	310.09	J/molxK	584.63	Joback Method

cpg	299.07	J/molxK	553.00	Joback Method
cpg	287.52	J/molxK	521.37	Joback Method
cpg	340.02	J/molxK	679.53	Joback Method
dvisc	0.0003165	Paxs	489.74	Joback Method
dvisc	0.0004145	Paxs	452.25	Joback Method
dvisc	0.0005701	Paxs	414.75	Joback Method
dvisc	0.0008352	Paxs	377.26	Joback Method
dvisc	0.0013314	Paxs	339.77	Joback Method
dvisc	0.0023823	Paxs	302.27	Joback Method
dvisc	0.0050267	Paxs	264.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1540358&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1540358&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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