

# Quinoline, 1,2,3,4-tetrahydro-2-methyl-

<b>Other names:</b>	Quinaldine, 1,2,3,4-tetrahydro- 1,2,3,4-Tetrahydro-2-methylquinoline 1,2,3,4-Tetrahydro-2-methylquinoline 1,2,3,4-Tetrahydroquinaldine 2-Methyl-1,2,3,4-tetrahydroquinoline
<b>Inchi:</b>	InChI=1S/C10H13N/c1-8-6-7-9-4-2-3-5-10(9)11-8/h2-5,8,11H,6-7H2,1H3
<b>InchiKey:</b>	JZICUKPOZUKZLL-UHFFFAOYSA-N
<b>Formula:</b>	C10H13N
<b>SMILES:</b>	CC1CCc2ccccc2N1
<b>Mol. weight [g/mol]:</b>	147.22
<b>CAS:</b>	1780-19-4

## Physical Properties

Property code	Value	Unit	Source
gf	272.46	kJ/mol	Joback Method
hf	79.78	kJ/mol	Joback Method
hfus	20.93	kJ/mol	Joback Method
hvap	47.64	kJ/mol	Joback Method
log10ws	-2.71		Crippen Method
logp	2.433		Crippen Method
mcvol	127.120	ml/mol	McGowan Method
pc	3534.66	kPa	Joback Method
rinpol	1406.00		NIST Webbook
tb	519.42	K	Joback Method
tc	757.27	K	Joback Method
tf	360.85	K	Joback Method
vc	0.473	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	284.82	J/mol×K	519.42	Joback Method
cpg	301.68	J/mol×K	559.06	Joback Method
cpg	317.42	J/mol×K	598.70	Joback Method

cpg	332.11	J/mol×K	638.35	Joback Method
cpg	345.79	J/mol×K	677.99	Joback Method
cpg	358.52	J/mol×K	717.63	Joback Method
cpg	370.35	J/mol×K	757.27	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1780194&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1780194&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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