

# Glutaric acid, 4-chloro-3-methylphenyl tetradecyl ester

Inchi:	InChI=1S/C26H41ClO4/c1-3-4-5-6-7-8-9-10-11-12-13-14-20-30-25(28)16-15-17-26(29)3
InchiKey:	WOMCMLAVBSYKGF-UHFFFAOYSA-N
Formula:	C26H41ClO4
SMILES:	CCCCCCCCCCCCCOC(=O)CCCC(=O)Oc1ccc(Cl)c(C)c1
Mol. weight [g/mol]:	453.05

## Physical Properties

Property code	Value	Unit	Source
gf	-218.58	kJ/mol	Joback Method
hf	-871.72	kJ/mol	Joback Method
hfus	66.13	kJ/mol	Joback Method
hvap	99.77	kJ/mol	Joback Method
log10ws	-8.92		Crippen Method
logp	7.968		Crippen Method
mcvol	380.560	ml/mol	McGowan Method
pc	886.83	kPa	Joback Method
rinpola	3374.00		NIST Webbook
tb	1020.93	K	Joback Method
tc	1251.86	K	Joback Method
tf	608.48	K	Joback Method
vc	1.480	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1268.95	J/molxK	1020.93	Joback Method
cpg	1334.34	J/molxK	1213.37	Joback Method
cpg	1324.34	J/molxK	1174.88	Joback Method
cpg	1312.85	J/molxK	1136.39	Joback Method
cpg	1299.83	J/molxK	1097.91	Joback Method
cpg	1285.21	J/molxK	1059.42	Joback Method
cpg	1342.90	J/molxK	1251.86	Joback Method
dvisc	0.0000196	Paxs	1020.93	Joback Method
dvisc	0.0000253	Paxs	952.19	Joback Method

dvisc	0.0000339	Paxs	883.45	Joback Method
dvisc	0.0000478	Paxs	814.71	Joback Method
dvisc	0.0000718	Paxs	745.96	Joback Method
dvisc	0.0001172	Paxs	677.22	Joback Method
dvisc	0.0002135	Paxs	608.48	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358804&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358804&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/52-731-2/Glutaric-acid-4-chloro-3-methylphenyl-tetradecyl-ester.pdf>

Generated by Cheméo on 2024-04-29 08:02:49.492284528 +0000 UTC m=+16667018.412861844.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.