

# 4,4'-Dichlorobenzophenone

<b>Other names:</b>	Methanone, bis(4-chlorophenyl)- Benzophenone, 4,4'-dichloro- p,p'-Dichlorobenzophenone Bis(4-chlorophenyl) ketone DBP DCBP Di(p-Chlorophenyl)ketone DCB USAF DO-4 p-Dichlorobenzophenone DBP (degradation product) NSC 8787 p,p-Dichlorobenzophenone Dichloro-4,4'-dibenzophenone
<b>Inchi:</b>	InChI=1S/C13H8Cl2O/c14-11-5-1-9(2-6-11)13(16)10-3-7-12(15)8-4-10/h1-8H
<b>InchiKey:</b>	OKISUZLXOYGIFP-UHFFFAOYSA-N
<b>Formula:</b>	C13H8Cl2O
<b>SMILES:</b>	O=C(c1ccc(Cl)cc1)c1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	251.11
<b>CAS:</b>	90-98-2

## Physical Properties

Property code	Value	Unit	Source
gf	111.36	kJ/mol	Joback Method
hf	-5.59	kJ/mol	Joback Method
hfus	26.72	kJ/mol	Joback Method
hsub	117.50 ± 0.30	kJ/mol	NIST Webbook
hvap	65.92	kJ/mol	Joback Method
log10ws	-4.80		Crippen Method
logp	4.224		Crippen Method
mcvol	172.560	ml/mol	McGowan Method
pc	2937.70	kPa	Joback Method
rinpol	2014.00		NIST Webbook
rinpol	2018.00		NIST Webbook
rinpol	1947.00		NIST Webbook
rinpol	1979.00		NIST Webbook
tb	626.20	K	NIST Webbook

tc	948.38	K	Joback Method
tf	420.00 ± 0.20	K	NIST Webbook
vc	0.651	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	387.24	J/mol×K	688.89	Joback Method
cpg	399.33	J/mol×K	732.14	Joback Method
cpg	410.32	J/mol×K	775.39	Joback Method
cpg	420.28	J/mol×K	818.64	Joback Method
cpg	429.27	J/mol×K	861.88	Joback Method
cpg	437.38	J/mol×K	905.13	Joback Method
cpg	444.68	J/mol×K	948.38	Joback Method
dvisc	0.0005279	Paxs	512.24	Joback Method
dvisc	0.0007755	Paxs	468.08	Joback Method
dvisc	0.0012341	Paxs	423.92	Joback Method
dvisc	0.0003820	Paxs	556.41	Joback Method
dvisc	0.0002899	Paxs	600.57	Joback Method
dvisc	0.0002285	Paxs	644.73	Joback Method
dvisc	0.0001857	Paxs	688.89	Joback Method
hfust	30.12	kJ/mol	420.00	NIST Webbook
hfust	30.12	kJ/mol	420.00	NIST Webbook
hfust	21.65	kJ/mol	338.40	NIST Webbook
hsubt	114.50 ± 0.30	kJ/mol	358.00	NIST Webbook

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C90982&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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