

Octane, 2-iodo-

Other names:	2-Iodooctane 2-Octyl iodide sec-Octyl Iodide
Inchi:	InChI=1S/C8H17I/c1-3-4-5-6-7-8(2)9/h8H,3-7H2,1-2H3
InchiKey:	XFLOGTUFKZCFTK-UHFFFAOYSA-N
Formula:	C8H17I
SMILES:	CCCCCCC(C)I
Mol. weight [g/mol]:	240.12
CAS:	557-36-8

Physical Properties

Property code	Value	Unit	Source
gf	72.16	kJ/mol	Joback Method
hf	-136.86	kJ/mol	Joback Method
hfus	17.36	kJ/mol	Joback Method
hvap	42.39	kJ/mol	Joback Method
log10ws	-4.23		Crippen Method
logp	3.780		Crippen Method
mcvol	149.400	ml/mol	McGowan Method
pc	2500.00	kPa	Joback Method
rinpol	1119.00		NIST Webbook
rinpol	1124.00		NIST Webbook
tb	475.14	K	Joback Method
tc	675.39	K	Joback Method
tf	222.98	K	Joback Method
vc	0.566	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	291.15	J/mol×K	475.14	Joback Method
cpg	353.06	J/mol×K	642.01	Joback Method
cpg	341.94	J/mol×K	608.64	Joback Method
cpg	330.22	J/mol×K	575.26	Joback Method

cpg	317.86	J/molxK	541.89	Joback Method
cpg	304.85	J/molxK	508.51	Joback Method
cpg	363.61	J/molxK	675.39	Joback Method
dvisc	0.0002962	Paxs	475.14	Joback Method
dvisc	0.0003989	Paxs	433.11	Joback Method
dvisc	0.0005725	Paxs	391.09	Joback Method
dvisc	0.0008966	Paxs	349.06	Joback Method
dvisc	0.0015876	Paxs	307.03	Joback Method
dvisc	0.0033695	Paxs	265.01	Joback Method
dvisc	0.0094969	Paxs	222.98	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.46081e+01
Coeff. B	-4.09643e+03
Coeff. C	-7.54660e+01
Temperature range (K), min.	361.52
Temperature range (K), max.	516.10

Sources

The Yaws Handbook of Vapor Pressure:
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C557368&Units=SI>

Legend

cpg: Ideal gas heat capacity
dvisc: Dynamic viscosity
gf: Standard Gibbs free energy of formation

hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
r inpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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