

# Propanoic acid, dodecyl ester

<b>Other names:</b>	Dodecyl propionate propionic acid, dodecyl ester
<b>Inchi:</b>	InChI=1S/C15H30O2/c1-3-5-6-7-8-9-10-11-12-13-14-17-15(16)4-2/h3-14H2,1-2H3
<b>InchiKey:</b>	FVGJPCFYGPKBKJ-UHFFFAOYSA-N
<b>Formula:</b>	C15H30O2
<b>SMILES:</b>	CCCCCCCCCCCCOC(=O)CC
<b>Mol. weight [g/mol]:</b>	242.40
<b>CAS:</b>	6221-93-8

## Physical Properties

Property code	Value	Unit	Source
gf	-158.50	kJ/mol	Joback Method
hf	-597.73	kJ/mol	Joback Method
hfus	37.39	kJ/mol	Joback Method
hvap	58.14	kJ/mol	Joback Method
log10ws	-4.96		Crippen Method
logp	4.861		Crippen Method
mcvol	229.650	ml/mol	McGowan Method
pc	1465.73	kPa	Joback Method
rinpol	1676.00		NIST Webbook
rinpol	1672.00		NIST Webbook
rinpol	1679.00		NIST Webbook
rinpol	1680.00		NIST Webbook
rinpol	1679.00		NIST Webbook
rinpol	1682.00		NIST Webbook
rinpol	1677.00		NIST Webbook
rinpol	1676.00		NIST Webbook
rinpol	1672.00		NIST Webbook
rinpol	1686.00		NIST Webbook
ripol	1995.00		NIST Webbook
ripol	1960.00		NIST Webbook
ripol	1959.00		NIST Webbook
ripol	1953.00		NIST Webbook
ripol	1953.00		NIST Webbook
ripol	1911.00		NIST Webbook
ripol	1930.00		NIST Webbook
ripol	1979.00		NIST Webbook

tb	618.89	K	Joback Method
tc	786.41	K	Joback Method
tf	330.97	K	Joback Method
vc	0.899	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	620.74	J/mol×K	618.89	Joback Method
cpg	638.14	J/mol×K	646.81	Joback Method
cpg	654.83	J/mol×K	674.73	Joback Method
cpg	670.81	J/mol×K	702.65	Joback Method
cpg	686.10	J/mol×K	730.57	Joback Method
cpg	700.70	J/mol×K	758.49	Joback Method
cpg	714.65	J/mol×K	786.41	Joback Method
dvisc	0.0025922	Paxs	330.97	Joback Method
dvisc	0.0011647	Paxs	378.96	Joback Method
dvisc	0.0006264	Paxs	426.94	Joback Method
dvisc	0.0003819	Paxs	474.93	Joback Method
dvisc	0.0002549	Paxs	522.92	Joback Method
dvisc	0.0001822	Paxs	570.90	Joback Method
dvisc	0.0001371	Paxs	618.89	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6221938&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6221938&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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