

# 4-Imidazolidinone, 2-thioxo-

<b>Other names:</b>	2-Thioguidanthion 2-Thioxo-4-imidazolinone 2-thiohydantoin 2-thioxo-4-imidazolidinone Glycine thiohydantoin Hydantoin, 2-thio- Thiohydantoin USAF BE-25
<b>Inchi:</b>	InChI=1S/C3H4N2OS/c6-2-1-4-3(7)5-2/h1H2,(H2,4,5,6,7)
<b>InchiKey:</b>	UGWULZWUXSCWPX-UHFFFAOYSA-N
<b>Formula:</b>	C3H4N2OS
<b>SMILES:</b>	O=C1CNC(=S)N1
<b>Mol. weight [g/mol]:</b>	116.14
<b>CAS:</b>	503-87-7

## Physical Properties

Property code	Value	Unit	Source
gf	162.32	kJ/mol	Joback Method
hf	28.59	kJ/mol	Joback Method
hfus	21.11	kJ/mol	Joback Method
hvap	48.08	kJ/mol	Joback Method
log10ws	-0.47		Crippen Method
logp	-1.009		Crippen Method
mcvol	75.850	ml/mol	McGowan Method
pc	7640.96	kPa	Joback Method
tb	525.55	K	Joback Method
tc	792.42	K	Joback Method
tf	506.10	K	Experimental and computational study of the energetics of hydantoin and 2-thiohydantoin
vc	0.265	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	140.87	J/mol×K	525.55	Joback Method
cpg	148.91	J/mol×K	570.03	Joback Method
cpg	156.55	J/mol×K	614.51	Joback Method
cpg	163.78	J/mol×K	658.99	Joback Method
cpg	170.59	J/mol×K	703.46	Joback Method
cpg	176.97	J/mol×K	747.94	Joback Method
cpg	182.89	J/mol×K	792.42	Joback Method
cps	137.05	J/mol×K	298.15	Evaluation of sublimation enthalpy by thermogravimetry: Analysis of the diffusion effects in the case of methyl and phenyl substituted hydantoins

## Sources

Experimental and computational study of the energetics of hydantoin and Evaluation of sublimation enthalpy by thermogravimetry: Analysis of the diffusion effects in the case of methyl and phenyl substituted hydantoins: McGowan Method:	<a href="https://www.doi.org/10.1016/j.jct.2012.10.010">https://www.doi.org/10.1016/j.jct.2012.10.010</a>
Experimental and computational study of the energetics of hydantoin and Evaluation of sublimation enthalpy by thermogravimetry: Analysis of the diffusion effects in the case of methyl and phenyl substituted hydantoins: McGowan Method:	<a href="https://www.doi.org/10.1016/j.tca.2017.06.024">https://www.doi.org/10.1016/j.tca.2017.06.024</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C503877&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C503877&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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