

# p-Benzoquinone, 2,3,5,6-tetrachloro-

<b>Other names:</b>	.alpha.-chloranil 1,2,4,5-tetrachlorobenzoquinone 1,4-Benzoquinone, 2,3,5,6-tetrachloro- 1,4-benzoquinone, tetrachloro- 2,3,5,6-Tetrachloro-1,4-benzoquinone 2,3,5,6-Tetrachloro-2,5-cyclohexadiene-1,4-dione 2,3,5,6-Tetrachloro-p-benzoquinone 2,3,5,6-Tetrachlorobenzoquinone 2,3,5,6-Tetrachloroquinone 2,5-Cyclohexadiene-1,4-dione, 2,3,5,6-tetrachloro- Chloranil Coversan Dow Seed Disinfectant No. 5 ENT 3,797 ENT 3797 G-25804 G-444E Geigy-444e Khloranil NSC 8432 Psorisan Quinone Tetrachloride Reranil Spergon Spergon I Spergon Technical Tetrachloro-1,4-benzoquinone Tetrachloro-p-benzoquinone Tetrachloro-p-quinone Tetrachlorobenzoquinone Tetrachloroparabenzoquinone Tetrachloroquinone Tetrachlorquinone Vulklor p-Chloranil «alpha»-Chloranil Â«alphaÂ»-Chloranil
<b>Inchi:</b>	InChI=1S/C6Cl4O2/c7-1-2(8)6(12)4(10)3(9)5(1)11
<b>InchiKey:</b>	UGNWTBMOAKPKBL-UHFFFAOYSA-N
<b>Formula:</b>	C6Cl4O2

**SMILES:** O=C1C(Cl)=C(Cl)C(=O)C(Cl)=C1Cl  
**Mol. weight [g/mol]:** 245.88  
**CAS:** 118-75-2

## Physical Properties

Property code	Value	Unit	Source
chs	-2171.10 ± 8.40	kJ/mol	NIST Webbook
chs	-2156.00	kJ/mol	NIST Webbook
chs	-2166.00	kJ/mol	NIST Webbook
ea	2.45 ± 0.26	eV	NIST Webbook
ea	2.76 ± 0.20	eV	NIST Webbook
ea	2.75 ± 0.05	eV	NIST Webbook
ea	2.78 ± 0.10	eV	NIST Webbook
gf	-239.70	kJ/mol	Joback Method
hf	-361.19	kJ/mol	Joback Method
hfs	-298.00	kJ/mol	NIST Webbook
hfus	18.76	kJ/mol	Joback Method
hsub	98.74	kJ/mol	NIST Webbook
hvap	58.95	kJ/mol	Joback Method
ie	9.90 ± 0.05	eV	NIST Webbook
ie	9.74	eV	NIST Webbook
log10ws	-3.09		Crippen Method
logp	2.517		Crippen Method
mcvol	128.040	ml/mol	McGowan Method
pc	3801.00	kPa	Joback Method
ss	258.40	J/mol×K	NIST Webbook
tb	664.50	K	Joback Method
tc	935.65	K	Joback Method
tf	568.13 ± 0.20	K	NIST Webbook
vc	0.487	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	256.77	J/mol×K	890.46	Joback Method
cpg	226.43	J/mol×K	664.50	Joback Method
cpg	233.75	J/mol×K	709.69	Joback Method

cpg	240.55	J/mol×K	754.88	Joback Method
cpg	246.72	J/mol×K	800.07	Joback Method
cpg	252.16	J/mol×K	845.26	Joback Method
cpg	260.45	J/mol×K	935.65	Joback Method
cps	193.80	J/mol×K	298.15	NIST Webbook
hfust	30.87	kJ/mol	567.20	NIST Webbook
hfust	30.87	kJ/mol	567.20	NIST Webbook
hsubt	98.70 ± 8.30	kJ/mol	344.50	NIST Webbook
hvapt	88.50	kJ/mol	389.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.57544e+01
Coeff. B	-1.19044e+04
Temperature range (K), min.	467.45
Temperature range (K), max.	582.32

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Spectroscopic and ultrasonic studies on the molecular interaction of certain aromatic amines with p-chloranil:</b>	<a href="https://www.doi.org/10.1016/j.fluid.2013.09.026">https://www.doi.org/10.1016/j.fluid.2013.09.026</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C118752&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C118752&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity

<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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