

trans-3-methoxycinnamic acid

Other names:	2-Propenoic acid, 3-(3-methoxyphenyl)- 3-(3-Methoxyphenyl)acrylic acid Cinnamic acid, m-methoxy- Propenoic acid, 3-(3-methoxyphenyl)-, trans- m-Methoxycinnamic acid trans-3-Methoxycinnamic acid
Inchi:	InChI=1S/C10H10O3/c1-13-9-4-2-3-8(7-9)5-6-10(11)12/h2-7H,1H3,(H,11,12)
InchiKey:	LZPNXAULYJPXEH-AATRIKPKSA-N
Formula:	C10H10O3
SMILES:	COc1cccc(C=CC(=O)O)c1
Mol. weight [g/mol]:	178.18
CAS:	17570-26-2

Physical Properties

Property code	Value	Unit	Source
hf	-386.41	kJ/mol	Cheméo Relay (1.0)
hfus	22.39	kJ/mol	Joback Method
hsub	124.00 ± 0.90	kJ/mol	NIST Webbook
hvap	96.96	kJ/mol	Cheméo Relay (1.0)
ie	8.43	eV	Cheméo Relay (1.0)
log10ws	-2.50		Cheméo Relay (1.0)
logp	1.793		Crippen Method
mcvol	137.010	ml/mol	McGowan Method
tb	593.04	K	Cheméo Relay (1.0)
tc	788.32	K	Cheméo Relay (1.0)
tf	372.84	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	328.56	J/mol×K	632.49	Joback Method
cpg	338.85	J/mol×K	667.04	Joback Method
cpg	348.51	J/mol×K	701.59	Joback Method
cpg	357.56	J/mol×K	736.14	Joback Method

cpg	366.02	J/mol×K	770.69	Joback Method
cpg	373.93	J/mol×K	805.24	Joback Method
cpg	381.31	J/mol×K	839.79	Joback Method
dvisc	0.0023160	Pa×s	369.30	Joback Method
dvisc	0.0008973	Pa×s	413.17	Joback Method
dvisc	0.0004170	Pa×s	457.03	Joback Method
dvisc	0.0002217	Pa×s	500.89	Joback Method
dvisc	0.0001304	Pa×s	544.76	Joback Method
dvisc	0.0000831	Pa×s	588.62	Joback Method
dvisc	0.0000563	Pa×s	632.49	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):	
Cheméo Relay (1.0, beta):	
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C17570262&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hsub:	Enthalpy of sublimation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point

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