

# Benzenamine, 2-(trifluoromethyl)-

<b>Other names:</b>	o-Toluidine, «alpha», «alpha», «alpha»-trifluoro- «alpha», «alpha», «alpha»-Trifluoro-o-toluidine o-(Trifluoromethyl)aniline o-Aminobenzotrifluoride 2-(Trifluoromethyl)aniline 2-(Trifluoromethyl)benzenamine 2-Aminobenzotrifluoride 2-Trifluoromethylaniline UN 2942 2-Trifluoromethyl-phenylamine NSC 10336 o-(Trifluoromethyl)phenylamine
<b>Inchi:</b>	InChI=1S/C7H6F3N/c8-7(9,10)5-3-1-2-4-6(5)11/h1-4H,11H2
<b>InchiKey:</b>	VBLXCTYLWZJBKA-UHFFFAOYSA-N
<b>Formula:</b>	C7H6F3N
<b>SMILES:</b>	Nc1ccccc1C(F)(F)F
<b>Mol. weight [g/mol]:</b>	161.12
<b>CAS:</b>	88-17-5

## Physical Properties

Property code	Value	Unit	Source
gf	-404.30	kJ/mol	Joback Method
hf	-526.04	kJ/mol	Joback Method
hfus	14.56	kJ/mol	Joback Method
hvap	41.01	kJ/mol	Joback Method
log10ws	-2.21		Crippen Method
logp	2.288		Crippen Method
mcvol	101.020	ml/mol	McGowan Method
pc	3749.97	kPa	Joback Method
rinpola	955.50		NIST Webbook
tb	458.33	K	Joback Method
tc	664.73	K	Joback Method
tf	295.04	K	Joback Method
vc	0.392	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	216.15	J/mol×K	458.33	Joback Method
cpg	226.94	J/mol×K	492.73	Joback Method
cpg	236.95	J/mol×K	527.13	Joback Method
cpg	246.21	J/mol×K	561.53	Joback Method
cpg	254.78	J/mol×K	595.93	Joback Method
cpg	262.69	J/mol×K	630.33	Joback Method
cpg	269.98	J/mol×K	664.73	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	339.00 ± 1.00	K	2.00	NIST Webbook
tbrp	341.20	K	2.00	NIST Webbook
tbrp	340.50 ± 0.50	K	2.00	NIST Webbook

# Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C88175&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C88175&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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