

# Dimethylmalonic acid, 1-phenyl-2-(cyclohex-2-enyl)ethyl propyl ester

Inchi:	InChI=1S/C22H30O4/c1-4-15-25-20(23)22(2,3)21(24)26-19(18-13-9-6-10-14-18)16-17-1
InchiKey:	QEDJNZDPFQOXKW-UHFFFAOYSA-N
Formula:	C22H30O4
SMILES:	CCCOC(=O)C(C)(C)C(=O)OC(CC1C=CCCC1)c1ccccc1
Mol. weight [g/mol]:	358.47

## Physical Properties

Property code	Value	Unit	Source
gf	-166.26	kJ/mol	Joback Method
hf	-652.41	kJ/mol	Joback Method
hfus	34.47	kJ/mol	Joback Method
hvap	84.19	kJ/mol	Joback Method
log10ws	-5.58		Crippen Method
logp	4.997		Crippen Method
mcvol	296.800	ml/mol	McGowan Method
pc	1435.89	kPa	Joback Method
rinpol	2315.00		NIST Webbook
tb	897.06	K	Joback Method
tc	1124.04	K	Joback Method
tf	504.00	K	Joback Method
vc	1.109	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	964.66	J/molxK	897.06	Joback Method
cpg	981.38	J/molxK	934.89	Joback Method
cpg	996.57	J/molxK	972.72	Joback Method
cpg	1010.32	J/molxK	1010.55	Joback Method
cpg	1022.69	J/molxK	1048.38	Joback Method
cpg	1033.76	J/molxK	1086.21	Joback Method
cpg	1043.62	J/molxK	1124.04	Joback Method
dvisc	0.0006527	Paxs	504.00	Joback Method
dvisc	0.0002964	Paxs	569.51	Joback Method

dvisc	0.0001584	Paxs	635.02	Joback Method
dvisc	0.0000952	Paxs	700.53	Joback Method
dvisc	0.0000624	Paxs	766.04	Joback Method
dvisc	0.0000437	Paxs	831.55	Joback Method
dvisc	0.0000323	Paxs	897.06	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361869&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361869&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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