

3-Butenoic acid

Other names:	Acetic acid, ethenyl- CH ₂ =CHCH ₂ COOH Vinylacetic acid but-3-enoic acid «beta»-Butenoic acid Â«betaÂ»-Butenoic acid
Inchi:	InChI=1S/C4H6O2/c1-2-3-4(5)6/h2H,1,3H2,(H,5,6)
InchiKey:	PVEOYINWKBTPIZ-UHFFFAOYSA-N
Formula:	C ₄ H ₆ O ₂
SMILES:	C=CCC(=O)O
Mol. weight [g/mol]:	86.09
CAS:	625-38-7

Physical Properties

Property code	Value	Unit	Source
gf	-195.10	kJ/mol	Joback Method
hf	-265.27	kJ/mol	Joback Method
hfus	10.52	kJ/mol	Joback Method
hvap	47.25	kJ/mol	Joback Method
ie	10.02	eV	NIST Webbook
ie	9.75	eV	NIST Webbook
log10ws	-0.45		Crippen Method
logp	0.647		Crippen Method
mcvol	70.360	ml/mol	McGowan Method
pc	4703.44 ± 100.00	kPa	NIST Webbook
ripol	1627.00		NIST Webbook
tb	436.20	K	NIST Webbook
tc	644.60 ± 3.00	K	NIST Webbook
tf	243.83	K	Joback Method
vc	0.266	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	130.77	J/molxK	433.65	Joback Method
cpg	136.62	J/molxK	463.22	Joback Method
cpg	142.21	J/molxK	492.78	Joback Method
cpg	147.53	J/molxK	522.35	Joback Method
cpg	152.61	J/molxK	551.92	Joback Method
cpg	157.44	J/molxK	581.48	Joback Method
cpg	162.04	J/molxK	611.05	Joback Method
dvisc	0.0295733	Paxs	243.83	Joback Method
dvisc	0.0087410	Paxs	275.47	Joback Method
dvisc	0.0033211	Paxs	307.10	Joback Method
dvisc	0.0015119	Paxs	338.74	Joback Method
dvisc	0.0007873	Paxs	370.38	Joback Method
dvisc	0.0004543	Paxs	402.01	Joback Method
dvisc	0.0002841	Paxs	433.65	Joback Method

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	342.70	K	1.60	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.37511e+01
Coeff. B	-2.95386e+03
Coeff. C	-1.19426e+02
Temperature range (K), min.	338.82
Temperature range (K), max.	469.43

Sources

The Yaws Handbook of Vapor Pressure:
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C625387&Units=SI

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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