

1,2-DIBROMO-4-NITROBENZENE

Inchi:	InChI=1S/C6H3Br2NO2/c7-5-2-1-4(9(10)11)3-6(5)8/h1-3H
InchiKey:	DLLDRYLYVHKDKK-UHFFFAOYSA-N
Formula:	C6H3Br2NO2
SMILES:	O=[N+]([O-])c1ccc(Br)c(Br)c1
Mol. weight [g/mol]:	280.90

Physical Properties

Property code	Value	Unit	Source
hf	96.30	kJ/mol	Cheméo Relay (1.0)
hfus	26.49	kJ/mol	Joback Method
hvap	78.62	kJ/mol	Cheméo Relay (1.0)
ie	9.76	eV	Cheméo Relay (1.0)
log10ws	-4.11		Cheméo Relay (1.0)
logp	3.120		Crippen Method
mcvol	124.060	ml/mol	McGowan Method
tb	567.29	K	Cheméo Relay (1.0)
tf	376.53	K	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	229.96	J/mol×K	657.48	Joback Method
cpg	237.11	J/mol×K	704.89	Joback Method
cpg	243.55	J/mol×K	752.31	Joback Method
cpg	249.34	J/mol×K	799.72	Joback Method
cpg	254.59	J/mol×K	847.14	Joback Method
cpg	259.35	J/mol×K	894.55	Joback Method
cpg	263.72	J/mol×K	941.96	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Cheméo Relay (1.0):	
Cheméo Relay (1.0, beta):	
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C5411507&Units=SI
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
tb:	Normal Boiling Point Temperature
tf:	Normal melting (fusion) point

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