

# Glutaric acid, pentyl 2,4,6-trichlorophenyl ester

Inchi:	InChI=1S/C16H19Cl3O4/c1-2-3-4-8-22-14(20)6-5-7-15(21)23-16-12(18)9-11(17)10-13(16)
InchiKey:	CEBIBRYSGQZPHR-UHFFFAOYSA-N
Formula:	C16H19Cl3O4
SMILES:	CCCCCOC(=O)CCCC(=O)Oc1c(Cl)cc(Cl)cc1Cl
Mol. weight [g/mol]:	381.68

## Physical Properties

Property code	Value	Unit	Source
gf	-336.27	kJ/mol	Joback Method
hf	-708.27	kJ/mol	Joback Method
hfus	48.23	kJ/mol	Joback Method
hvap	86.94	kJ/mol	Joback Method
log10ws	-6.05		Crippen Method
logp	5.456		Crippen Method
mcvol	264.140	ml/mol	McGowan Method
pc	1616.77	kPa	Joback Method
rinpol	2537.00		NIST Webbook
rinpol	2537.00		NIST Webbook
tb	871.97	K	Joback Method
tc	1087.15	K	Joback Method
tf	568.14	K	Joback Method
vc	1.018	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	720.38	J/molxK	871.97	Joback Method
cpg	768.04	J/molxK	1051.29	Joback Method
cpg	760.53	J/molxK	1015.42	Joback Method
cpg	752.01	J/molxK	979.56	Joback Method
cpg	742.49	J/molxK	943.70	Joback Method
cpg	731.95	J/molxK	907.83	Joback Method
cpg	774.56	J/molxK	1087.15	Joback Method
dvisc	0.0000638	Paxs	871.97	Joback Method

dvisc	0.0000786	Paxs	821.33	Joback Method
dvisc	0.0000995	Paxs	770.69	Joback Method
dvisc	0.0001302	Paxs	720.05	Joback Method
dvisc	0.0001774	Paxs	669.42	Joback Method
dvisc	0.0002544	Paxs	618.78	Joback Method
dvisc	0.0003890	Paxs	568.14	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358982&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358982&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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