

# Thiobenzoic acid, S-n-butyl ester

<b>Other names:</b>	n-Butylthio benzoate Butyl thiobenzoate Benzenecarbothioic acid, S-butyl ester Benzoic acid, thio-, S-butyl ester Thiobenzoic acid S-butyl ester S-Butyl thiobenzoate Butyl thiolbenzoate S-Butyl benzenecarbothioate
<b>Inchi:</b>	InChI=1S/C11H14OS/c1-2-3-9-13-11(12)10-7-5-4-6-8-10/h4-8H,2-3,9H2,1H3
<b>InchiKey:</b>	KXXZPIFZTPWOJJ-UHFFFAOYSA-N
<b>Formula:</b>	C11H14OS
<b>SMILES:</b>	CCCCSC(=O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	194.29
<b>CAS:</b>	7269-35-4

## Physical Properties

Property code	Value	Unit	Source
gf	58.35	kJ/mol	Joback Method
hf	-104.55	kJ/mol	Joback Method
hfus	24.02	kJ/mol	Joback Method
hvap	55.92	kJ/mol	Joback Method
log10ws	-3.77		Crippen Method
logp	3.360		Crippen Method
mcvol	160.010	ml/mol	McGowan Method
pc	2881.21	kPa	Joback Method
rinpol	1640.00		NIST Webbook
tb	600.41	K	Joback Method
tc	830.60	K	Joback Method
tf	324.48	K	Joback Method
vc	0.604	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	374.46	J/mol×K	600.41	Joback Method
cpg	389.18	J/mol×K	638.77	Joback Method
cpg	402.90	J/mol×K	677.14	Joback Method
cpg	415.66	J/mol×K	715.50	Joback Method
cpg	427.50	J/mol×K	753.87	Joback Method
cpg	438.45	J/mol×K	792.23	Joback Method
cpg	448.55	J/mol×K	830.60	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7269354&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7269354&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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