

# Thiophene, tetrabromo-

<b>Other names:</b>	2,3,4,5-tetrabromothiophene perbromothiophene tetrabromothiophene
<b>Inchi:</b>	InChI=1S/C4Br4S/c5-1-2(6)4(8)9-3(1)7
<b>InchiKey:</b>	AVPWUAFYDNQGNZ-UHFFFAOYSA-N
<b>Formula:</b>	C4Br4S
<b>SMILES:</b>	Brc1sc(Br)c(Br)c1Br
<b>Mol. weight [g/mol]:</b>	399.72
<b>CAS:</b>	3958-03-0

## Physical Properties

Property code	Value	Unit	Source
ie	8.53	eV	NIST Webbook
log10ws	-5.88		Crippen Method
logp	4.798		Crippen Method
mcvol	134.110	ml/mol	McGowan Method
tb	599.20	K	NIST Webbook
tf	391.75	K	Solubility and solution thermodynamics of 2,3,4,5-tetrabromothiophene in (ethanol + trichloromethane) binary solvent mixtures

## Sources

Measurement and correlation of the solubility of 2,3,4,5-tetrabromothiophene in different solvents: McGowan Method  
NIST Webbook:  
Crippen Method:  
Crippen Method:  
Solubility and solution thermodynamics of 2,3,4,5-tetrabromothiophene in (ethanol + tetrahydrofuran) binary solvent mixtures:

<https://www.doi.org/10.1016/j.jct.2012.06.005>

<https://www.doi.org/10.1016/j.jct.2013.05.023>

<http://link.springer.com/article/10.1007/BF02311772>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C3958030&Units=SI>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

<https://www.doi.org/10.1016/j.fluid.2013.11.023>

Solubility and solution thermodynamics of 2,3,4,5-tetrabromothiophene in (ethanol + trichloromethane) binary solvent mixtures:

# Legend

<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>tb:</b>	Normal Boiling Point Temperature
<b>tf:</b>	Normal melting (fusion) point

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